Problem Solving Class: Van Quark tot Biomaterie

Test Set: Numerical solutions to the Schrödinger equation of the hydrogen atom

The known wave functions for electrons in a hydrogen atom are (do not take Z=1, see below):

$$\psi_{n,l,m}(r,\theta,\phi) = \frac{u_{n,l}(r)}{r} Y_{l,m}(\theta,\phi)$$
$$u_{n,l}(r) = \sqrt{\frac{2Z}{na_0}} \sqrt{\frac{(n-l-1)!}{2n(n+l)!}} \exp(-\frac{Zr}{na_0}) \left(\frac{2Zr}{na_0}\right)^{l+1} L_{n-l-1}^{2l+1}(\frac{2Zr}{na_0})$$

Here L is a generalized Laguerre polynomial and Y is a spherical harmonic function.

Look up in the Mathematica library and find the functions: **LaguerreL** and **SphericalHarmonicY**, and use these functions in the following.

We found that Mathematica only finds solutions as long as *Z* remains defined as a variable; this is strange and we do not understand but use this hint to your advantage. (A bug ?).

- a) Write the wave function for a *ls* orbital (electronic wave function).
- b) Do the same for a *3p* orbital.
- c) Show that this *3p* gives a solution to the Schrödinger equation (by simply substituting).
- d) Write a general expression for an *np* wave function (with variable *n*, declared as an integer).

We now proceed in trying to calculate *intensities* of spectral lines. In general the intensity of a spectral line is given by the square of a matrix element, like in:

$$|\langle \psi_{f} | z | \psi_{i} \rangle|^{2}$$

Here this squared matrix element represents the intensity of a spectra line for a transition from the initial state $|i\rangle$ to a final state $|f\rangle$ where the external field is linearly polarized along the *z*-axis. Note that through a frame transformation this can e written as:

$$|\langle \psi_{f} | z | \psi_{i} \rangle|^{2} = |\langle \psi_{f} | r \cos(\theta) | \psi_{i} \rangle|^{2}$$

This is needed because the wave functions are defined in the coordinate system (r, θ , ϕ).

Selection Rules

Now write down an expression for a transition from state $|Is\rangle$ to state $|3p\rangle$. Since we are only interested in a selection rule you may ignore the radial part of the wave functions; focus on the angular part ! Make a calculation for the transition line intensity for three different possibilities. For the *Is* state there is only one option, namely the wave function:

 $\psi_{1,0,0}$

For the 3p state there are three possibilities however:

$$\psi_{3,1,1}, \psi_{3,1,0}$$
 and $\psi_{3,1,-1}$

Do the calculation for each of these three cases and find a selection rule for *m*-quantum numbers.

Polarization of the light

Now do the same calculation for the polarization vector (x+iy) instead of z.

 $|\langle \psi_{3p} | x + iy | \psi_{1s} \rangle|^2$

Now you find a different *m*-selection rule. Note that x+iy represents circularly polarized light.

Lyman series

Now we try to find a generalized expression for the spectral line intensity of the Lyman series. This is about the transitions: $ls \rightarrow np$. Use the knowledged from the previous section, while we use linearly polarized light (defined along the *z*-axis); then the quantum numbers are:

$$n_i = n$$
 $l_i = 1$ $m_i = 0$
 $n_f = 1$ $l_f = 0$ $m_f = 0$

We need to calculate the following integral:

$$<\psi_{1,0,0} \mid z \mid \psi_{n,1,0} > = <\psi_{1,0,0} \mid r \cos(\theta) \mid \psi_{n,1,0} >$$

This can be rewritten as:

$$<\psi_{1,0,0} | r \cos(\theta) | \psi_{n,1,0} >= \int d^{3}r (\psi_{1,0,0})^{*} r \cos(\theta) \psi_{n,1,0}$$
$$= \int_{0}^{2\pi\pi} \int_{0}^{\pi} Y_{0,0}(\theta,\varphi)^{*} \cos(\theta) Y_{1,0}(\theta,\varphi) \sin(\theta) d\theta d\varphi \int_{0}^{\infty} (\frac{u_{1,0}(r)}{r})^{*} r (\frac{u_{n,1}(r)}{r}) r^{2} dr$$

e) Use Mathematica and calculate the integral for the angular part.

Note that the intensity of a spectral line is given by the square of the matrix element. Neglect a proportionality factor:

$$I = |\langle \psi_{1,0,0} | z | \psi_{n,1,0} \rangle|^2 = |\langle \psi_{1,0,0} | r \cos(\theta) | \psi_{n,1,0} \rangle|^2$$

f) Calculate the *n*-dependence of this intensity. It should be possible to calculate this in algebraic form (with some difficulties); but if that does not turn out to be possible then attempt to obtain numerical solutions.

g) Approximate this *n*-dependence in the limit for large *n*.