Atoms and Molecules in the Universe the need for studying spectroscopy



А	O_2	759.370
В	O ₂	686.719
С	Нα	656.281
a	O ₂	627.661
D ₁	Na	589.592
D ₂	Na	588.995
D_3 (or d)	Не	587.5618



Solar spectrum with Fraunhofer lines and atmospheric water absorption band indicated.



Joseph von Fraunhofer (1787-1827)

First elements formed after Big Bang



Primordial nucleosynthesis of elements: only 3 minutes ! (limited lifetime of neutrons)

H, D, ⁴He, ³He (24%, ~all neutrons) ⁷Li

Abundance ratio is proof of Big Bang model



Heavy elements produced in death of stars: Supernovae



-massive stars (10 M_{sun}) lives only 10⁷ years

-hydrogen at inside is exhausted and is burned into helium

-core contracts under gravity and starts fusion of He into Carbon and Oxygen

-repeated pattern of core collaps and subsequent fusion

-last stage: Si burns into Fe (takes one week)

-gravitational collaps in 1s 10 km diameter, only neutrons

-potential energy causes a shock emission of energy (99% neutrinos)





Structure of diatoms





Poly-atomic molecules



$$KE_{restational} = \frac{1}{2}I\omega^2 = \frac{J^2}{2I}$$
$$KE_{rest} = \frac{1}{2}I_e\omega_e^2 + \frac{1}{2}I_b\omega_b^2 + \frac{1}{2}I_e\omega_e^2$$

in general 3 axes of rotation: 3 inertial moments

3 rotational quantum numbers:

 J, K_{a}, K_{c}

Microwave spectroscopy gives I-moments directly



Observation of molecules mostly by Radio-Astronomy



Westerbork Telescope Array



Only in atmospheric "window



"The Unidentified Infrared (UIR) Bands and the PAH-hypothesis



Infrared spectroscopic features observed, mostly from hot regions (dust exposed to UV radiation)



Interstellar Masers



Orion nebula

Hydroxyl radical - OH



Also: H₂O, SiO, Methanol masers



Molecules observed in Interstellar Space											
Number of Atoms											
2	3	4	5	6	7	8	9	10	11	12+	
H ₂ AIF AICI C ₂ CH CH' CN CO CO' CP CSi HCI KCI NH NO NS NaCI OH PN	C_3 C_2H C_2O C_2S CH_2 HCN HCO' HCO' HCO' HCO' HCO' HOC' H_2O H_2S HNC HNO MgCN MgNC N_2H' N_2O NaCN	$c-C_3H$ $I-C_3H$ C_3N C_3O C_3S C_2H_2 $CH_2D^{+}?$ HCCN $HCNH^+$ HNCO HNCS $HOCO^+$ H_2CO H_2CN H_2CS H_3O^+ NH_3 SiC_3 CH_3	C_5 C_4H C_4Si $I-C_3H_2$ $e-C_3H_2$ CH_2CN CH_4 HC_3N HC_2NC HCOOH H_2CHN H_2CHN H_2C_2O H_2NCN HNC_3 SiH_4 H_2COH^2	C_3H H_2C_4 C_2H_4 CH_3CN CH_3OH CH_3OH CH_3SH HC_3NH' HC_2CHO NH_2CHO C_3N	C ₀ H CH ₂ CHCN CH ₃ C ₂ H HC ₅ N HCOCH ₃ NH ₂ CH ₃ c-C ₂ H ₄ O	CH ₃ C ₃ N HCOOCH ₃ CH ₃ COOH? C ₇ H H ₂ C ₆	CH ₃ C ₄ H CH ₃ CH ₂ CN (CH ₃) ₂ O CH ₃ CH ₂ OH HC ₇ N C _g H	CH ₃ C ₅ N? (CH ₃) ₂ CO NH ₂ CH ₂ COOH	HC ₉ N	$C_{\theta}H_{\delta}$ H $C_{11}N$ PAHs $C_{\theta0}$ '?	
SO SO [†] SIN	SO ₂ c-SiC ₂				Detect	ion of a	mino ac	etonitrile	in Sgi	r B2(N)	
SiO SiS CS HF	CO ₂ NH ₂ H ₃ ⁺ H ₂ D ⁺									2008	





Search for Time Variation of the Fine Structure Constant

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Space-Time Variation of Physical Constants and Relativistic Corrections in Atoms

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Quasars give us the opportunity to look back in time; The *looking* is done by means of spectroscopy.









Atoms and clocks





frequency / time standard 9,192,631,770 Herz







Nobelprijs 2005

Frequency combs: extreme precision



J. Hall



T.W. Hänsch

"for their contributions to the development of laser-based precision spectroscopy including the optical frequency comb technique"



Future atomic clocks: in the optical domain

Ion traps?



Hg⁺, Al⁺, Sr⁺, Yb⁺? NIST, NPL, PTB

Optical lattices ?





