Lecture Course

Advanced Experimental Methods

W. Ubachs; part B

Detection of molecular species (with lasers)

Techniques

Issues

Direct absorption techniques Cavity Ring Down Cavity Enhanced Ionization detection (multi-photon) Laser-induced fluorescence Photo-acoustic technique Nonlinear optical techniques Optogalvanic technique

Resolution Spectroscopy vs quantitative Pressure Color (excitation level) Quantum state selectivity

Direct absorption

Beer's law, Beer-Lambert law

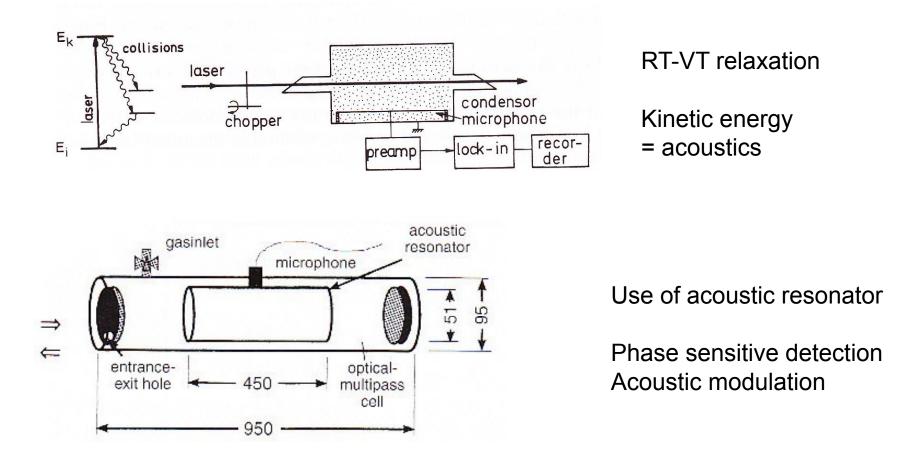
$$I_{v} = I_{v}^{0} \exp\left[-\sigma_{v} n l\right]$$

Exponent	$\sigma_{_{\scriptscriptstyle V}}$	cross section in [cm ²]
	n	density in [cm ⁻³]
	l	absorption length in [cm]
	nl	column density in [cm ⁻²]

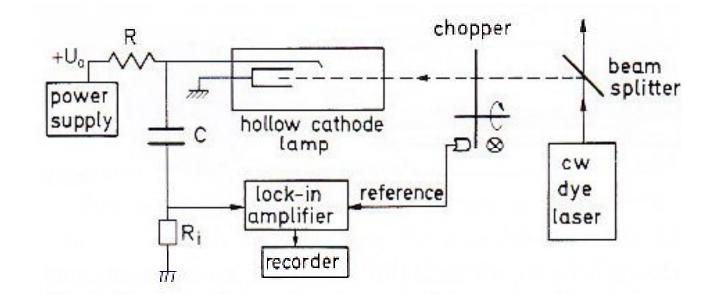
Problem:

- Measuring against a non-zero baseline

Photo-acoustic detection

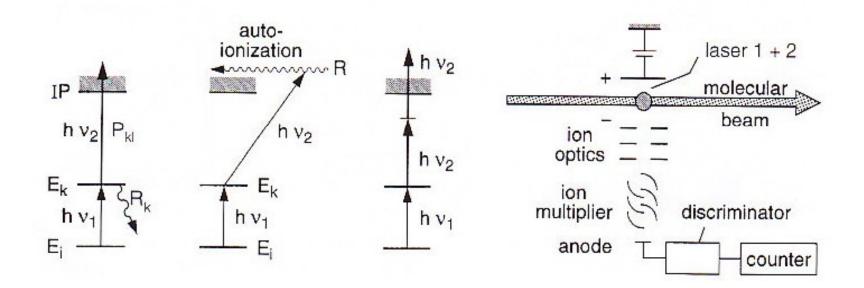


Optogalvanic detection



Principle: change the resistance over a plasma discharge by resonantly exciting atoms/molecules

Ionization detection

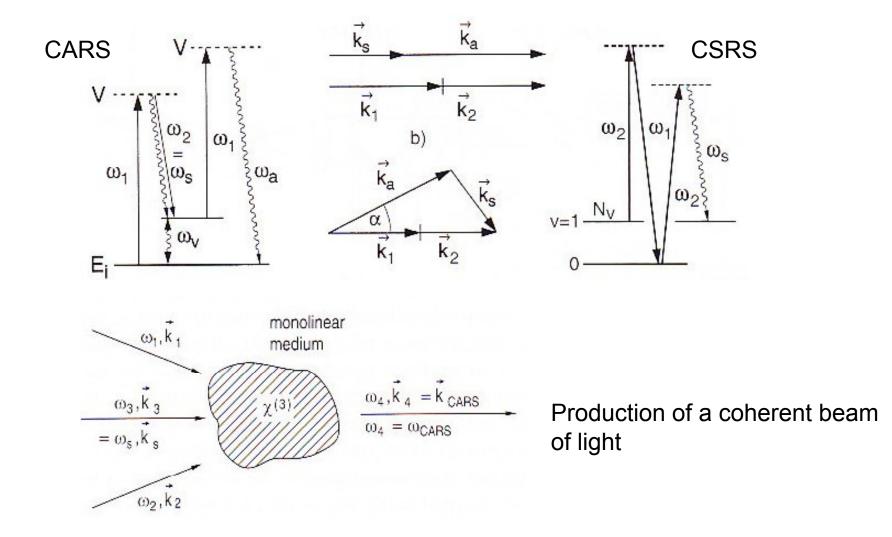


Various production schemes for ions

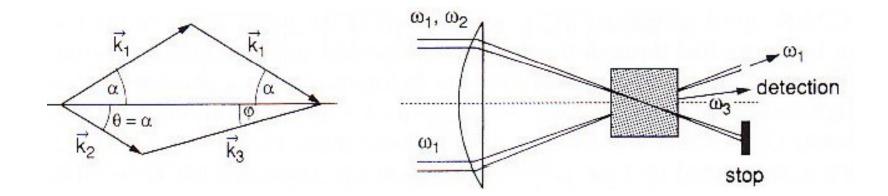
lons/charged particles sensitive detection

Time-of-flight mass spectrometry

CARS = Coherent Anti-Stokes Raman

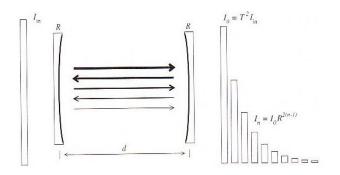


BOXCARS



Coherent beams in 3 dimensions

Cavity Ring-Down spectroscopy



CRD-analysis (assumption: no interference)

First pulse of light leaking out of the cavity:

$$I_0 = T^2 I_{in}$$

The nth pulse with intensity:

$$t = 2d(n-1)/c$$

 $I = I_{c} R^{2(n-1)} = I_{c} \rho^{2(n-1)\ln R}$

Will leak out of the cavity later:

$$u_n - 2a(n-1)/c$$

Smoothing over individual pulses: $I(t) = I_0 e^{-\left(\frac{c}{d}\right) |\ln R|t}$

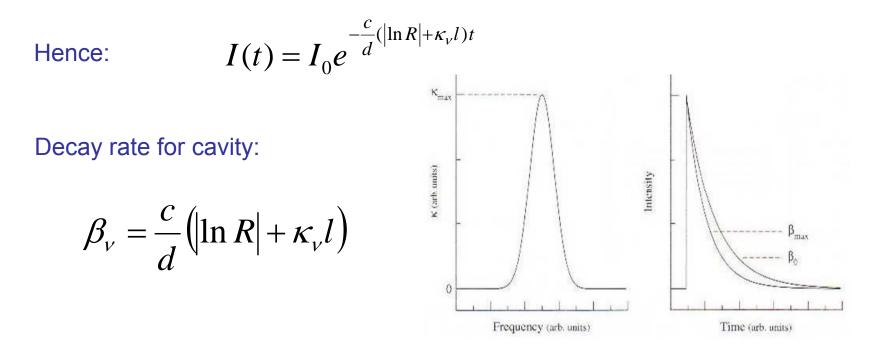
pulses.
$$I(l) - I_0 c$$

Hence for empty cavity: decay time: $\tau_0 = 1/\beta_0$ with: $\beta_0 = \frac{c|\ln R|}{r}$

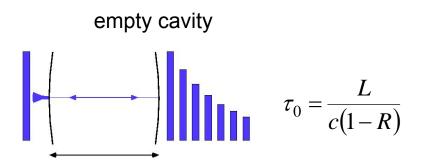
Cavity with absorbing species

Absorption coefficient : K_{ν}

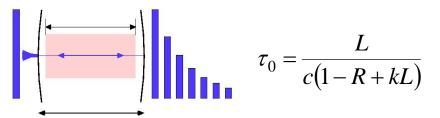
Cavity loss with absorption $I_n = I_0 R^{2(n-1)} e^{-2(n-1)\kappa_v} = I_0 e^{2(n-1)\ln R - 2(n-1)\kappa_v}$ for nth pulse:



CRD; the paradigm



cavity filled with absorbing gas



absorption coefficient; cross section

$$k = n\sigma = \frac{1}{c} \left(\frac{1}{\tau_0} - \frac{1}{\tau} \right)$$

What is it good for ?

- Spectroscopy (not extreme precision)
- Sensitive detection (within limits)
- Intensity quantitative (within severe limits/difficulty)
- Gases, to Solids, Surfaces, Liquids
- Wavelength range (limited – mirror quality)

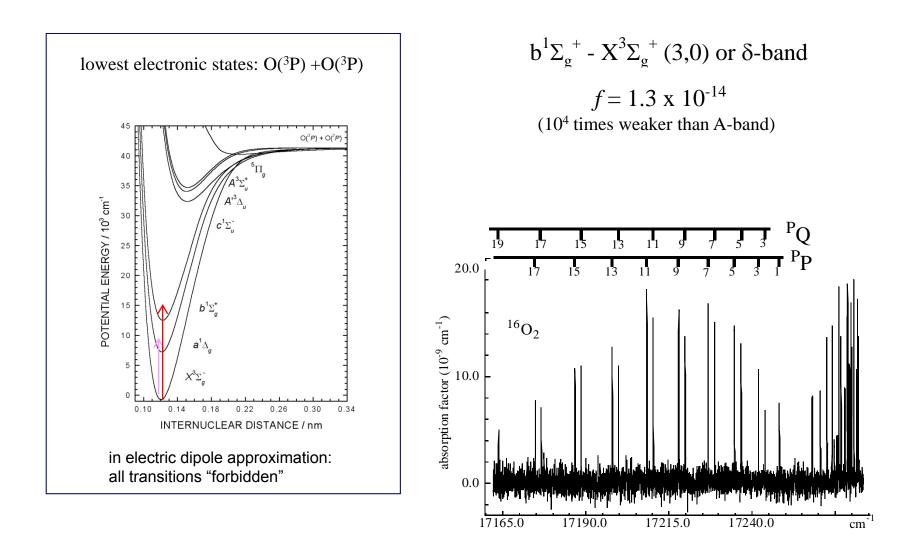
Advantages:

-No dependence on laser fluctuations -Extremely long effective path lengths -Easy to make "absolute"

L=1 m R=99.99%

 $\tau = 30\mu s$ Path=10 km

Weak transitions; in the benchmark CRD molecule: oxygen



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CRD in quantitative spectroscopy: Rayleigh Scattering



J.W. Strutt, 3rd Baron of Rayleigh

1899: full theory based on Electromagnetism

1918: Depolarization effects R.J. Strutt

1923: Depolarization and Cross section: King

1969: Full QM theory - Penney

No distinction of fine structure: Raman, Brillouin, Rayleigh wing, Cabannes single molecule cross section

$$\sigma(v) = \frac{24\pi^{3}v^{4}}{N^{2}} \frac{\left(n_{v}^{2}-1\right)^{2}}{\left(n_{v}^{2}+2\right)^{2}} F_{k(v)}$$

King factor: polarization effect

$$F_{k(\nu)} = \frac{6 + 3\rho_n(\nu)}{6 - 7\rho_n(\nu)} = \frac{3 + 6\rho_p(\nu)}{3 - 4\rho_p(\nu)} = 1 + 2\left(\frac{\gamma_{\nu}}{3\overline{\alpha_{\nu}}}\right)^2 > 1$$

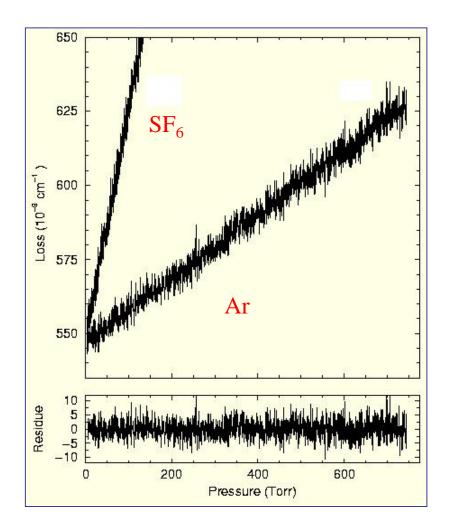
 ρ_n, ρ_p molecular depolarization,

$$\sigma(v) = \bar{\sigma}v^{4+\epsilon}$$

 \rightarrow Rayleigh scattering from the

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Rayleigh scattering: direct measurement

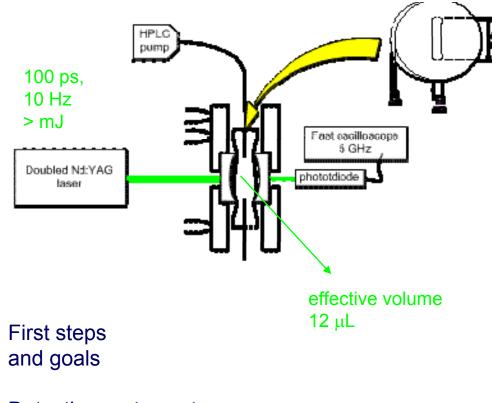


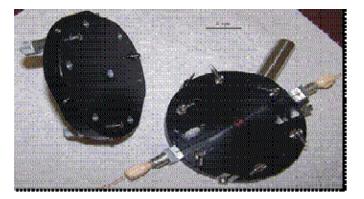
Cavity Loss: $\beta_v/c = |\ln R|/L + \sigma_v N$ $\sigma_v = \underline{\sigma} v^{4+\epsilon}$ for N₂: $\underline{\sigma}_{th} = 23.00 (0.23) 10^{-45}$ $\underline{\sigma}_{exp} = 22.94 (0.12) 10^{-45}$ for Ar: $\underline{\sigma}_{th} = 20.04 (0.05) 10^{-45}$ $\underline{\sigma}_{exp} = 19.89 (0.14) 10^{-45}$ for SF₆: $\underline{\sigma}_{th} = 183 (6) 10^{-45}$ $\underline{\sigma}_{exp} = 180 (6) 10^{-45}$

Theory – QM - ab initio: Oddershede/Svendsen $\sigma(N_2, 500 \text{ nm})$: 6.2 x 10⁻²⁷ cm²/molecule (10% off)

method: Naus and Ubachs, Opt. Lett. 25 (2000) 347

To the liquid phase: The combination CRD – HPLC in our approach

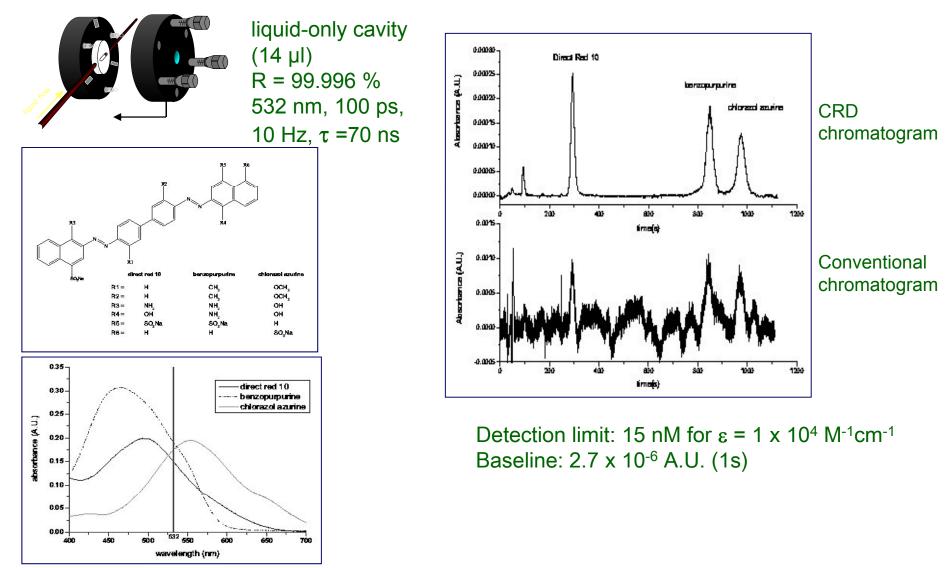




Philosophy: Liquid-Only cell

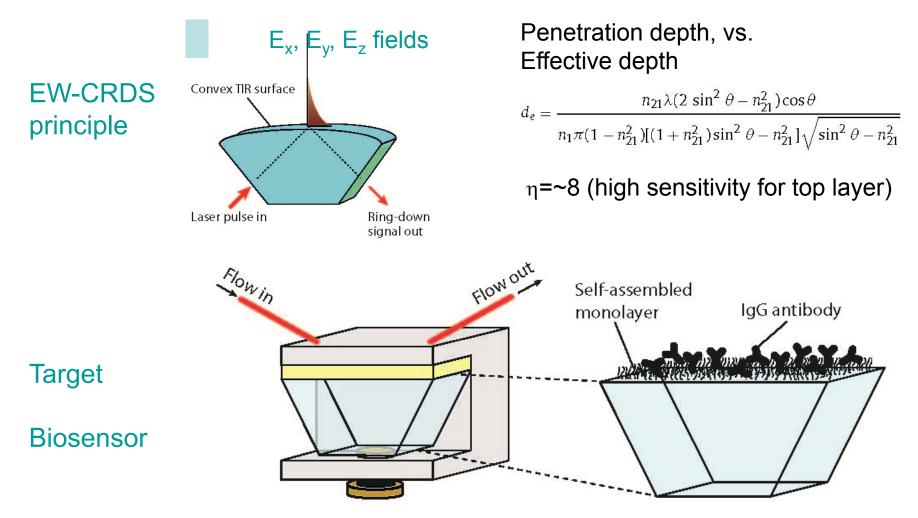
Detection, not spectroscopy Small volumes Universal wavelengths

Demonstration of online HPLC detection in Liquid-Only cell

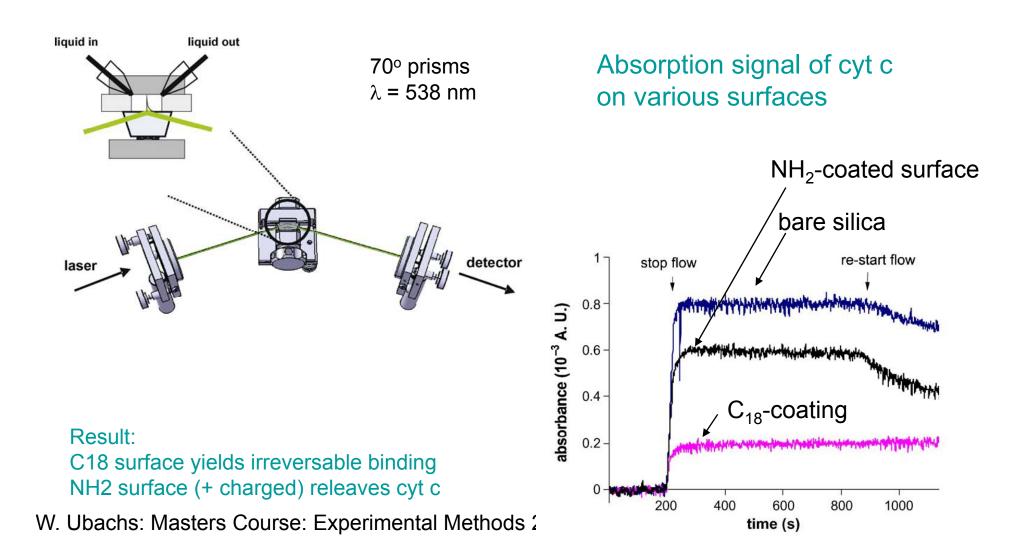


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Evanescent wave CRDS

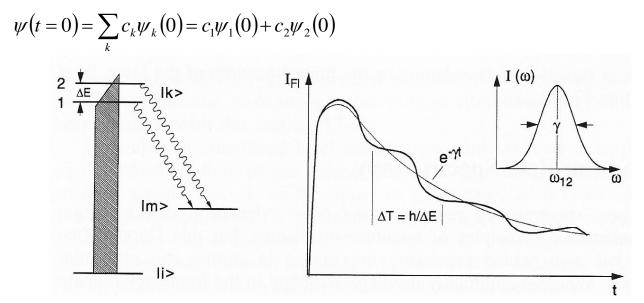


Toward a biosensor (cytochrome c as test molecule)



Quantum beat spectroscopy

Excitation of a coherent superposition of quantum states



Limitation →Natural lifetime (not laser pulse !!)

Temporal evolution of the superposition

$$\psi(t) = \sum_{k} c_k \psi_k(0) e^{-(i\omega_k + \gamma_k/2)t}$$

Total fluorescence emitted

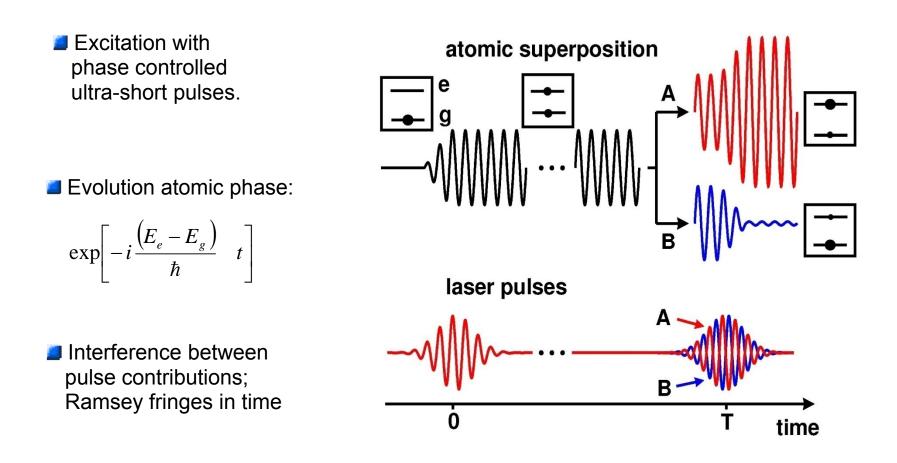
So
$$I(t) \propto e^{-\gamma t} (A + B \cos \omega_{21} t)$$

$$I(t) \propto \left| \langle \psi(t) | \varepsilon \cdot \mu | \psi_m \rangle \right|^2$$

with
$$A = c_1^2 |\langle \psi_1 | \varepsilon \cdot \mu | \psi_m \rangle|^2 + c_2^2 |\langle \psi_2 | \varepsilon \cdot \mu | \psi_m \rangle|^2$$

 $B = 2c_1c_2 |\langle \psi_1 | \varepsilon \cdot \mu | \psi_m \rangle| \cdot |\langle \psi_2 | \varepsilon \cdot \mu | \psi_m \rangle|$
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Direct frequency comb spectroscopy



Direct frequency comb spectroscopy

