Energy Transfer in Supramolecular Calix[4]arene – Perylene Bisimide Dye

Light Harvesting Building Blocks:

Resolving Loss Processes with Simultaneous Target Analysis

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Chemical structures of the calix[4]arene – perylene bisimide supramolecular systems

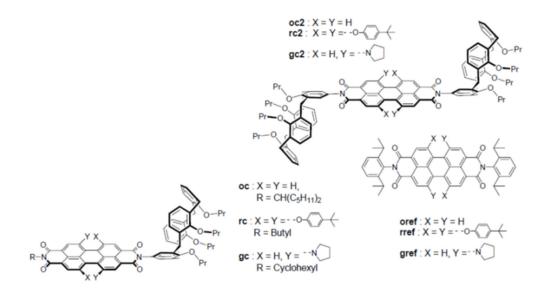


Figure S 1. Chemical structures of PBI–calix[4]arene conjugates and reference compounds studied. Figure adopted from [37]. **coc=oc2**, **crc=rc2**, **cgc=gc2**.

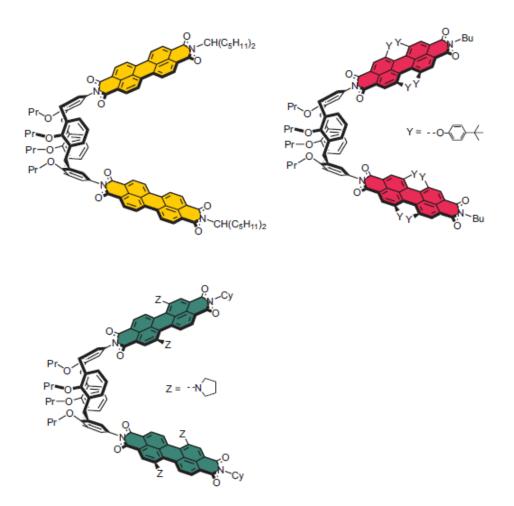


Figure S 2. Chemical structures of the bichromophoric supramolecular systems **oco**, **rcr** and **gcg**. Figure adopted from [37].

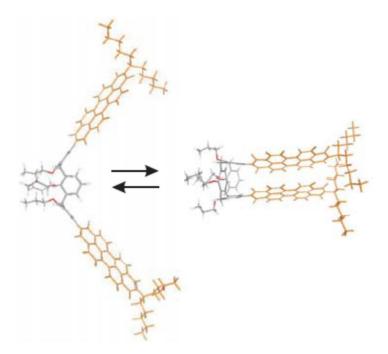


Figure S 3. Schematic representation of the equilibrium between non-stacked (left) and π -stacked (right) pinched cone conformation of the calix[4]arene unit upon substitution with two orange PBI units (for the chemical structure of **oco** see Figure S 2). Molecular structures obtained from force field calculations (Macromodel 8.0, potential MMFF). Color of the PBI chromophore is applied for clarity. Figure taken from [37].

Spectral evolution in oref, rref, gref, rc, crc, gc and cgc

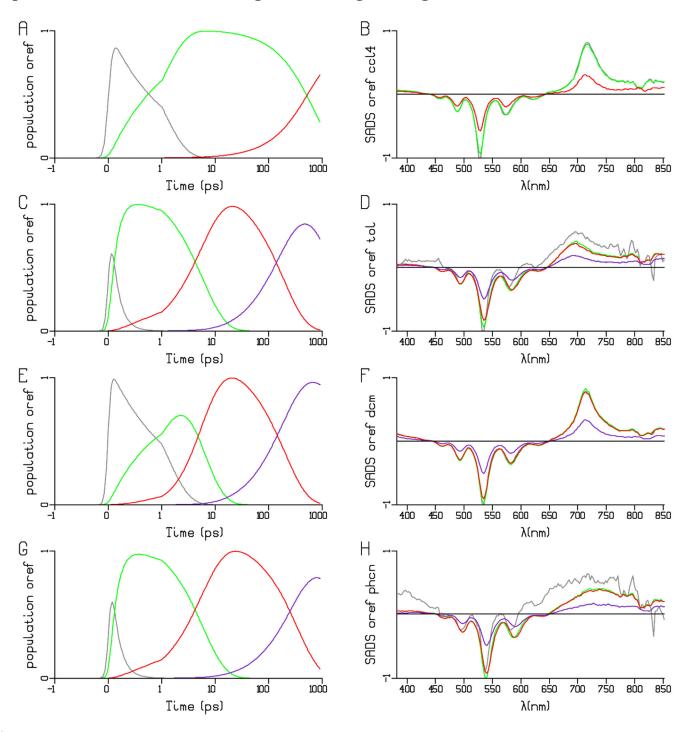


Figure S 4. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of reference compound **oref** in CCl4, toluene, CH₂Cl₂ and benzonitrile. Key: grey, **oref*FC**: excited **oref** chromophore in the Franck-Condon state; green, red, purple: successively relaxed **oref*** states. Lifetimes are collated in Table S 1. The SADS in the Supplementary figures have all been scaled to the maximum absolute value.

compound	solvent	τ1	τ2	τ3	τ4
oref	CCl4	1.08	713.7	5000	
	toluene	0.14	5.8	185	1759
	CH2Cl2	1.21	5.1	197	5000
	benzonitrile	0.14	5.8	337	2574
rref	methylcyclohexane	0.49	10.4	241	5000
	CCl4	0.75	14.3	344	5000
	toluene	0.40	8.3	222	4160
	CH2CI2	0.32	4.4	120	1147
	benzonitrile	0.53	14.0	255	4425
gref	toluene	0.25	3.2	290	4032
	CH2Cl2	0.14	1.4	220	2120
	benzonitrile	0.31	8.1	339	3163

Table S 1. Lifetimes (in ps) of the reference compounds in the different solvents.

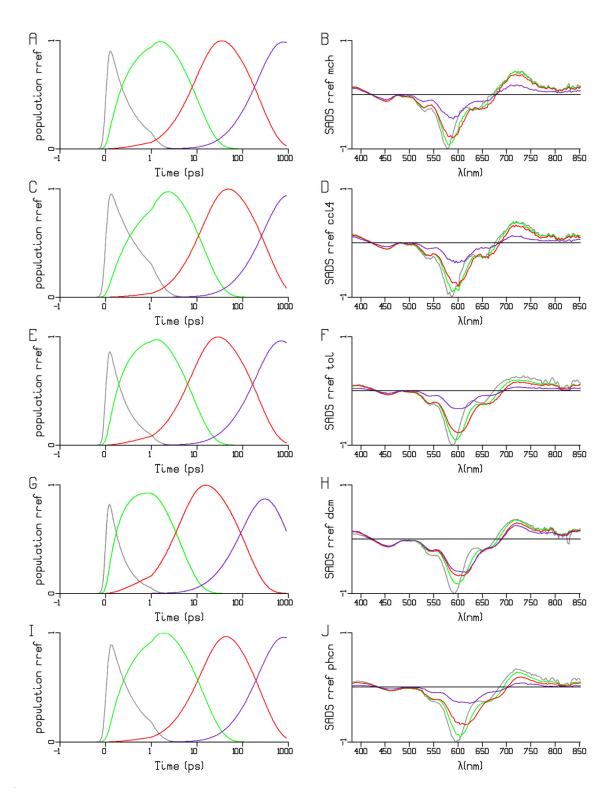


Figure S 5. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of reference compound **rref** in methylcyclohexane, CCl4, toluene, CH₂Cl₂ and benzonitrile. Key: grey, **rref*FC**: excited **rref** chromophore in the Franck-Condon state; green, red, purple: successively relaxed **rref*** states. Lifetimes are collated in Table S 1.

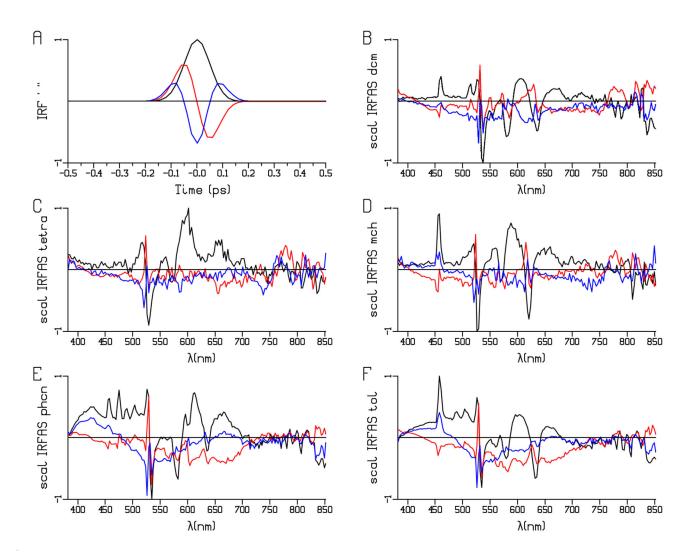


Figure S 6. Coherent artefact of the reference compound **rref** in CH₂Cl₂, CCl4, methylcyclohexane, benzonitrile and toluene. (A) 0th, 1st and 2nd derivative of the IRF (black, red and blue) which possessed a Full Width at Half Maximum of 119 fs. (B) scaled IRFAS in CH₂Cl₂. Scaling of the IRFAS is such that the product of the IRFAS and the IRF derivative is the contribution to the fit. Thus, the black IRFAS has the largest contribution to the fit. It shows large amplitudes straddling 530 nm, the excitation wavelength. In addition, Raman scattering is visible as negative peaks at 580 and 636 nm, and positive peaks at 460 and 512 nm. (C-F) scaled IRFAS in CCl4, methylcyclohexane, benzonitrile and toluene.

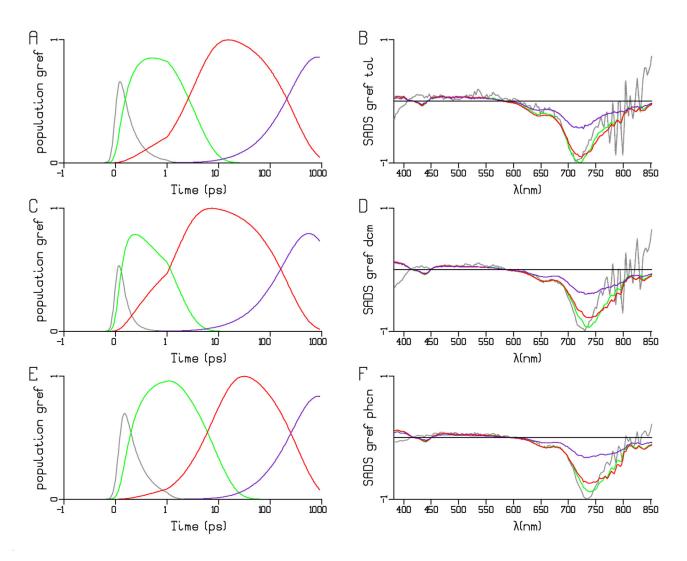


Figure S 7. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of reference compound **gref** in toluene, CH₂Cl₂ and benzonitrile. Key: grey, **gref*FC**: excited **gref** chromophore in the Franck-Condon state; green, red, purple: successively relaxed **gref*** states. Lifetimes are collated in Table S 1.

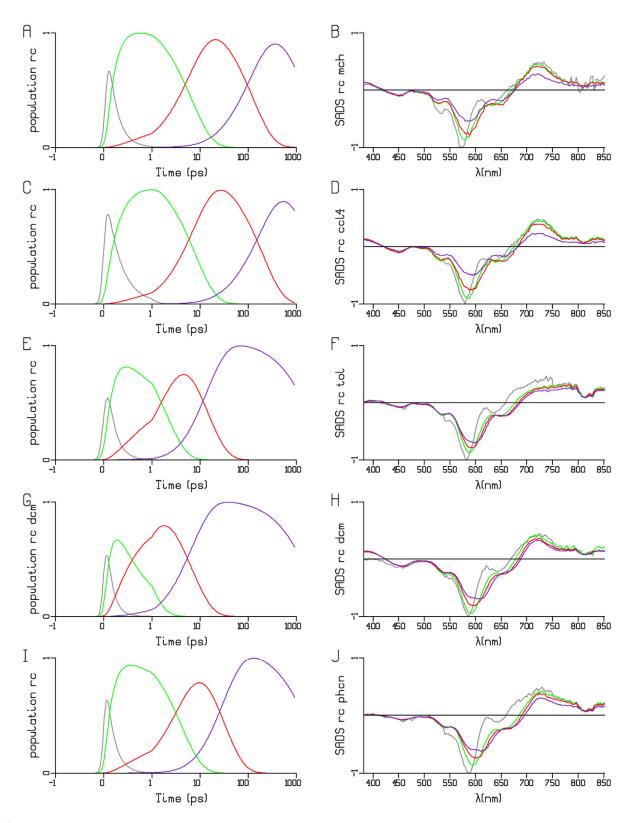


Figure S 8. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of **rc** and **crc** in methylcyclohexane, CCl4, toluene, CH₂Cl₂ and benzonitrile. Key: grey, green, red, purple: successively relaxed $\mathbf{r_1}^*$, $\mathbf{r_2}^*$, $\mathbf{r_3}^*$, $\mathbf{r_4}^*$. Lifetimes are collated in Table S 2.

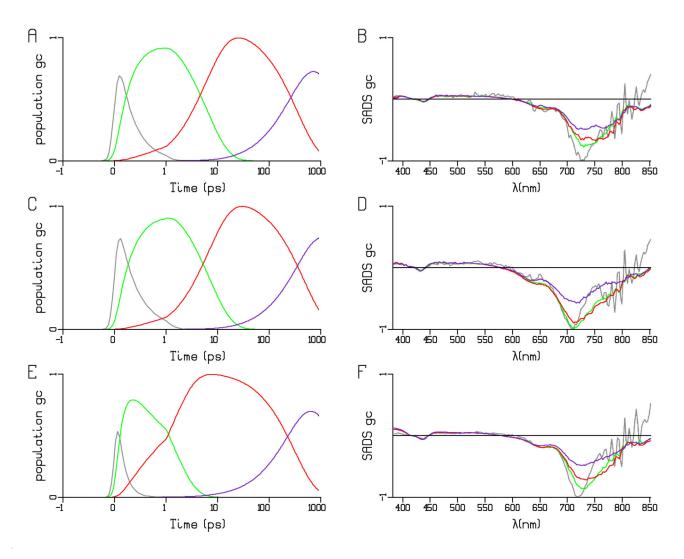
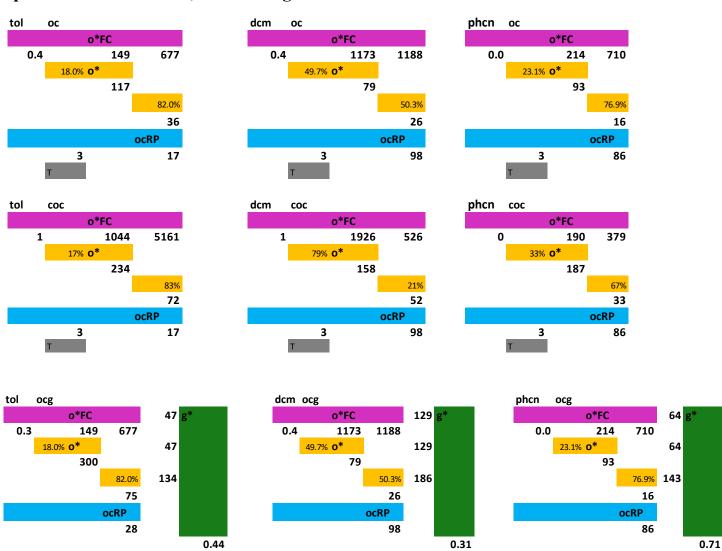


Figure S 9. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of **gc** and **cgc** in toluene, CH₂Cl₂ and benzonitrile. Key: grey, g*FC: excited **g** chromophore in the Franck-Condon state; green, red, purple: successively relaxed g* states. Lifetimes are collated in Table S 2.

compound	solvent	τ1	τ2	τ3	τ4
rc	methylcyclohexane	0.19	6.6	124	1721
	CCl4	0.30	7.9	198	2183
	toluene	0.17	1.9	13	1964
	CH2Cl2	0.13	0.6	7	1749
	benzonitrile	0.17	4.1	30	1805
gc	toluene	0.36	6.9	473	2829
	CH2CI2	0.13	1.4	308	1669
	benzonitrile	0.31	6.1	349	1813

Table S 2. Lifetimes (in ps) of the **rc** and **gc** complexes in the different solvents.



Spectral evolution in oc, coc and ocg

Figure S 10. Kinetic schemes used for the simultaneous target analysis of **oc**, **coc** and **ocg** in toluene, CH₂Cl₂ and benzonitrile. All rate constants in ns⁻¹. Key: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, two different conformations of **o*** with identical SADS; blue, **ocRP**: $\mathbf{c}^+\mathbf{o}^-$ radical ion pair; dark green: \mathbf{g}^* ; dark grey, T: triplet. Note that the RP formation and decay rates of **ocg** in toluene differ from those with **oc**. Further explanation in the caption of Figure 5.

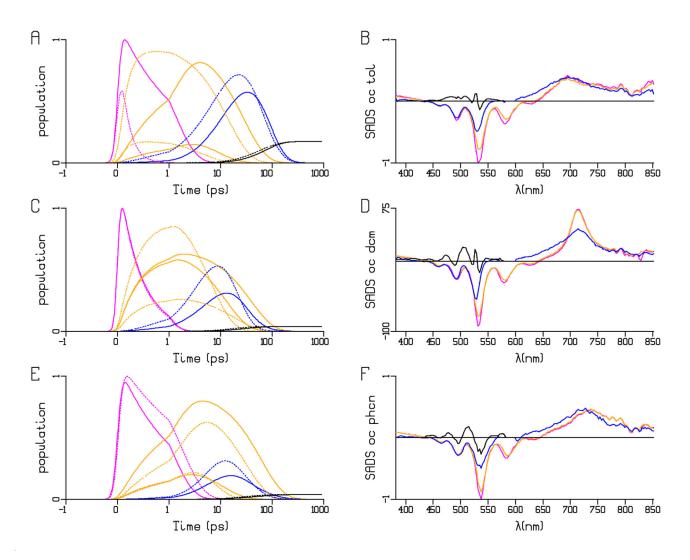


Figure S 11. Populations of the target kinetic schemes from Figure S 10 (A,C,E) and estimated SADS (B,D,F) of **oc** and **coc** in toluene, CH₂Cl₂ and benzonitrile, respectively. The **oc** and **coc** populations are depicted as solid and dotted lines, respectively. Key: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, **o*** relaxed; blue, **ocRP**: $c^+ \cdot o^-$ radical ion pair; dark grey, T: triplet.

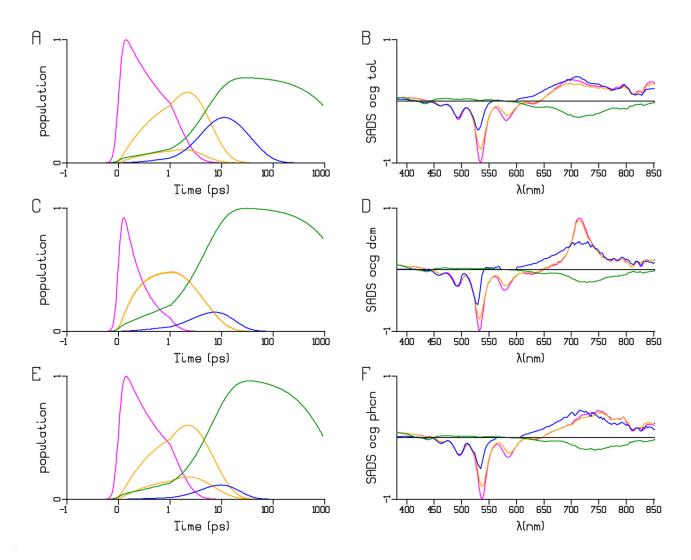


Figure S 12. Populations of the target kinetic schemes from Figure S 10 (A,C,E) and estimated SADS (B,D,F) of **ocg** in toluene, CH₂Cl₂ and benzonitrile, respectively. Key: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, **o*** relaxed; blue, **ocRP**: $c^+ o^-$ radical ion pair; dark green: **g***.

Spectral evolution in oco, rcr and gcg

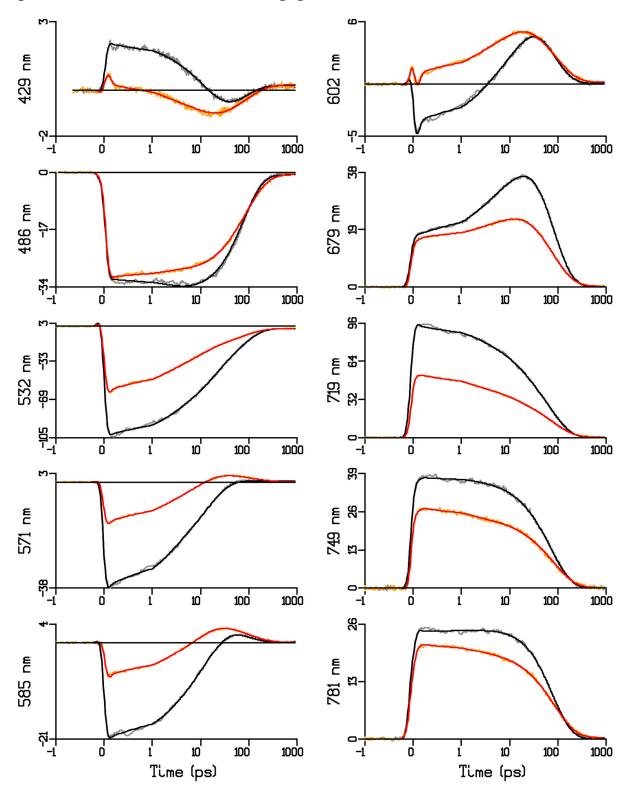


Figure S 13. Selected time traces in CCl₄ after excitation at 530 nm data (in mOD, grey, orange) and fit (black, red) of **oc** and **oco**, respectively. Wavelength is indicated in the ordinate label. Note that the time axis is linear until 1 ps (after the maximum of the IRF), and logarithmic thereafter.

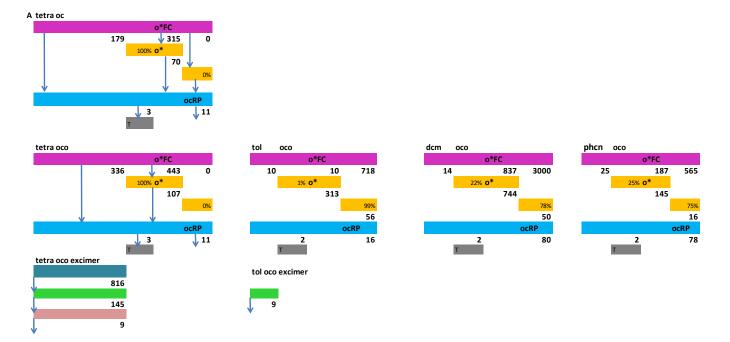


Figure S 14. Kinetic schemes used for the simultaneous target analysis of **oco** in CCl₄, toluene, CH₂Cl₂ and benzonitrile. All rate constants in ns⁻¹. Key nonstacked conformation: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, two different conformations of **o*** with identical SADS; blue, **ocRP**: $c^+ \cdot o^-$ radical ion pair; black, grey: triplet. Key π -stacked sandwich arrangement: turquoise, green, brown sequential evolution of the excimer state in CCl₄.

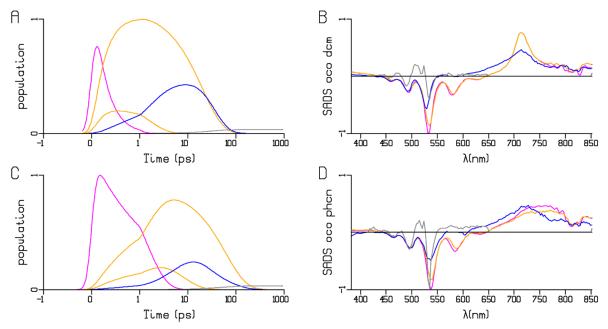


Figure S 15. Populations of the target kinetic schemes from (A,C) and estimated SADS (B,D) of **oco** in CH₂Cl₂ and benzonitrile, respectively. Key nonstacked conformation: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, **o*** relaxed; blue, **ocRP**: $c^+ \cdot o^-$ radical ion pair; grey: triplet.

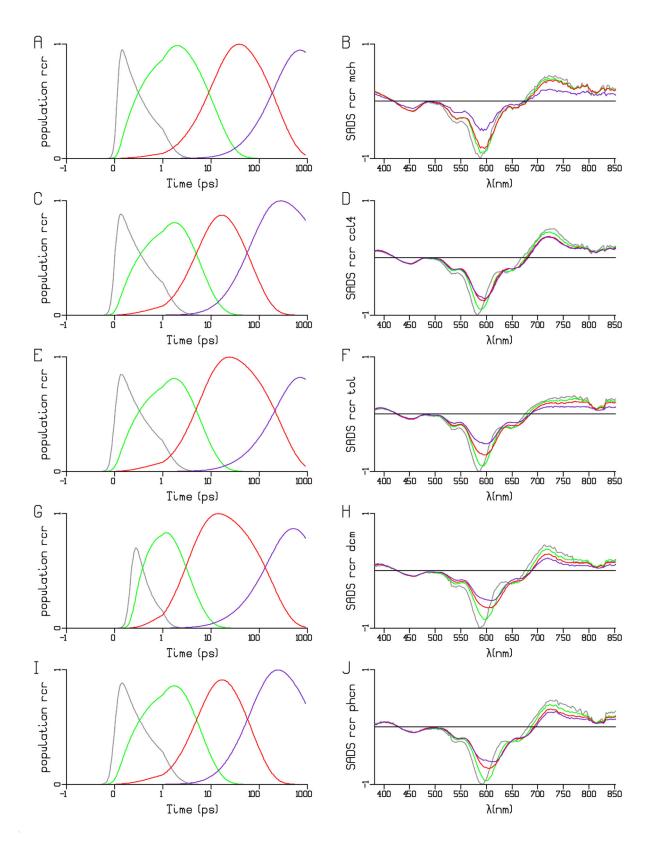


Figure S 16. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of **rcr** in methylcyclohexane, CCl4, toluene, CH₂Cl₂ and benzonitrile. Key: grey, $\mathbf{r}^*\mathbf{FC}$: excited \mathbf{r} chromophore in the Franck-Condon state; green, red, purple: successively relaxed \mathbf{r}^* states. Lifetimes are collated in Table S 3.

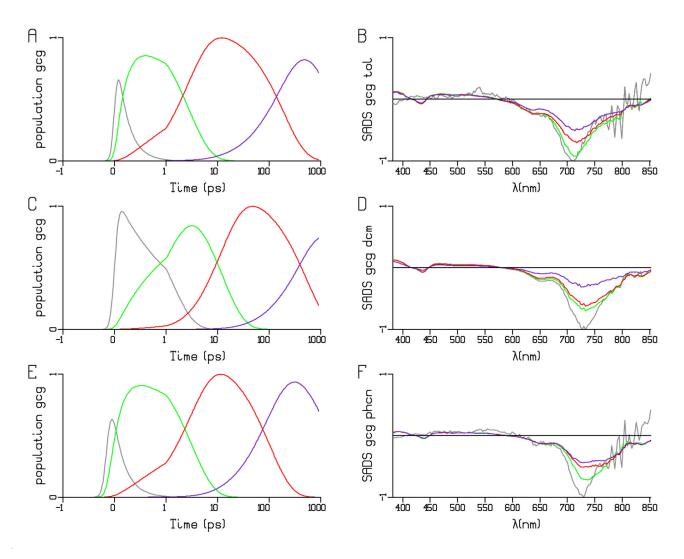


Figure S 17. Populations of a sequential kinetic scheme (A,C,E,G) and estimated SADS (B,D,F,H) of **gcg** in toluene, CH_2Cl_2 and benzonitrile. Key: grey, **g*FC**: excited **g** chromophore in the Franck-Condon state; green, red, purple: successively relaxed **g*** states. Lifetimes are collated in Table S 3.

compound	solvent	τ1	τ2	τ3	τ4
rcr	methylcyclohexane	0.62	11.7	245	3333
	CCl4	0.73	5.8	71	3025
	toluene	0.71	5.9	285	2533
	CH2CI2	0.33	3.3	182	2373
	benzonitrile	0.70	6.0	71	1731
gcg	toluene	0.21	2.7	189	1857
	CH2CI2	1.24	12.0	520	2912
	benzonitrile	0.22	3.1	102	1731

Table S 3. Lifetimes (in ps) of the rcr and gcg complexes in the different solvents

Spectral evolution in rc, rcg, rcgcr and gcrcg

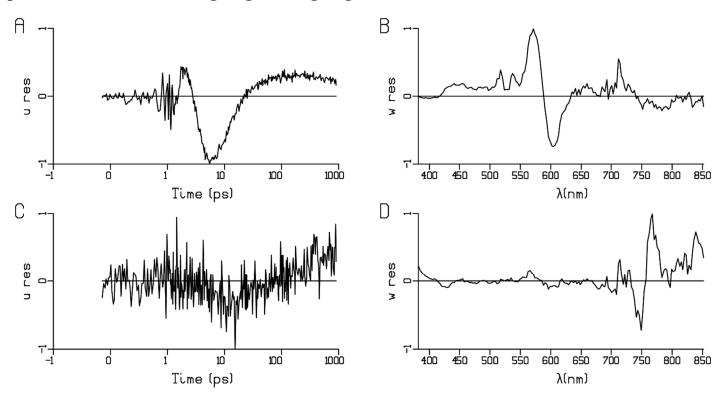


Figure S 18. First left and right singular vectors resulting from the singular value decomposition (SVD) of the residual matrix of **rcg** in CH₂Cl₂ resulting from a target analysis using a kinetic scheme without (A,B) or with (C,D) the **rcg** radical pair state. Note that panels (A,B) show large trends, whereas panels (C,D) show only small trends.

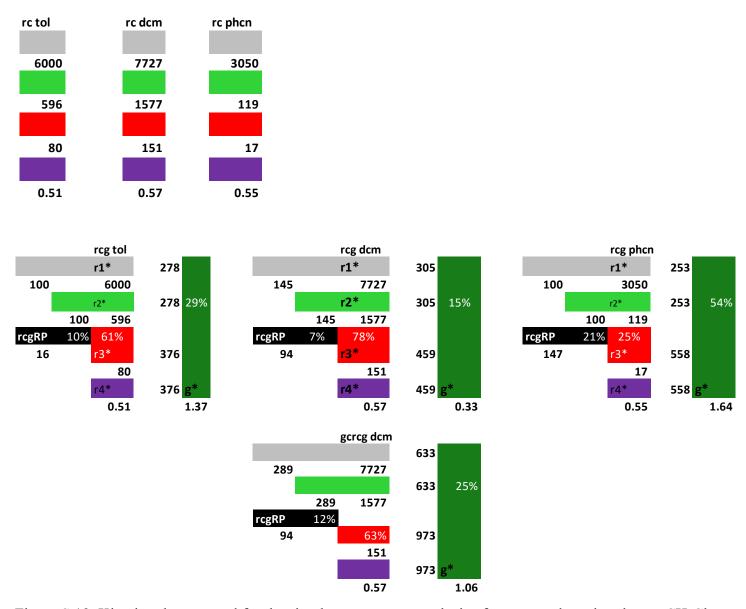


Figure S 19. Kinetic schemes used for the simultaneous target analysis of **rc**, **crc** and **rcg** in toluene, CH₂Cl₂ and benzonitrile, and for **gcrcg** in CH₂Cl₂. The kinetic scheme for **rcgcr** is identical to that of **rcg**, since there is only one **g** acceptor. All rate constants in ns⁻¹. Key: grey, green, red, purple: successively relaxed $\mathbf{r_1}^*$, $\mathbf{r_2}^*$, $\mathbf{r_3}^*$, $\mathbf{r_4}^*$; dark green: \mathbf{g}^* ; black, **rcgRP**: **rcg** radical pair. **rcgRP** is an important loss population formed from **rcg** (only from the $\mathbf{r_1}^*$ and $\mathbf{r_2}^*$ state). For further explanation cf. the arrows in Figure 9.

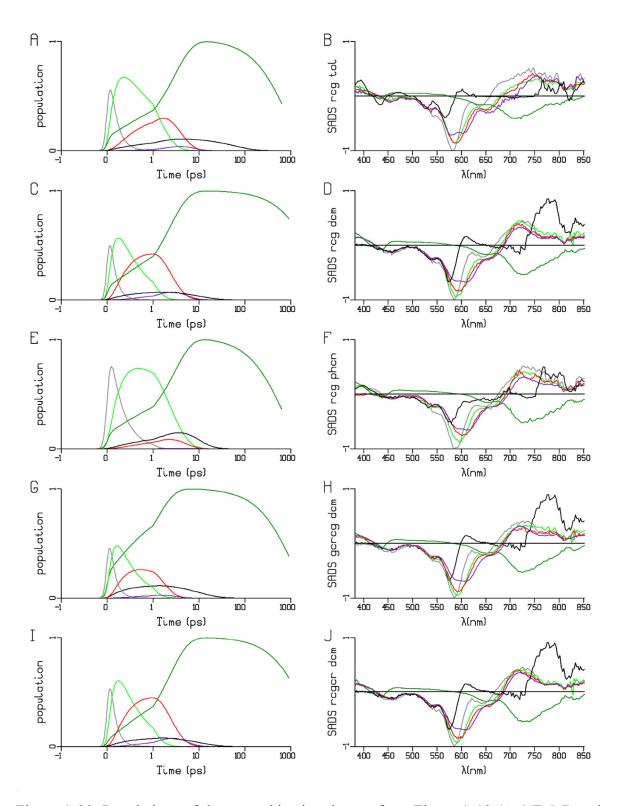


Figure S 20. Populations of the target kinetic schemes from Figure S 19 (A,C,E,G,I) and estimated SADS (B,D,F,H,J) of **rcg** in toluene, CH₂Cl₂ and benzonitrile, respectively, and **gcrcg** and **rcgcr** in CH₂Cl₂. Key: grey, green, red, purple: successively relaxed $\mathbf{r_1}^*$, $\mathbf{r_2}^*$, $\mathbf{r_3}^*$, $\mathbf{r_4}^*$; dark green: \mathbf{g}^* ; black, **rcgRP**: **rcg** radical pair. Note that the black SADS of rcg in toluene in (B) contains an additional bleach of \mathbf{g} near 430 nm.

Spectral evolution in ocr, rcocr and ocrco

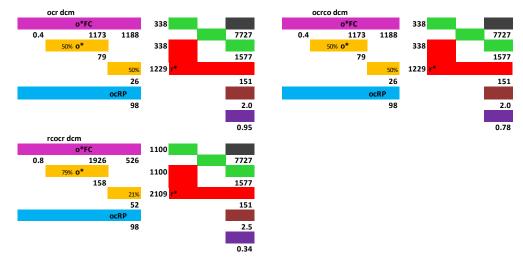


Figure S 21. Kinetic schemes used for the simultaneous target analysis of **ocr**, **rcocr** and **ocrco** in CH₂Cl₂. All rate constants in ns⁻¹. Key: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, two different conformations of **o*** with identical SADS; blue, **ocRP**: $c^+ \cdot o^-$ radical ion pair; grey, green, red, brown, purple: successively relaxed r_1^* , r_2^* , r_3^* , r_4^* , r_5^* .

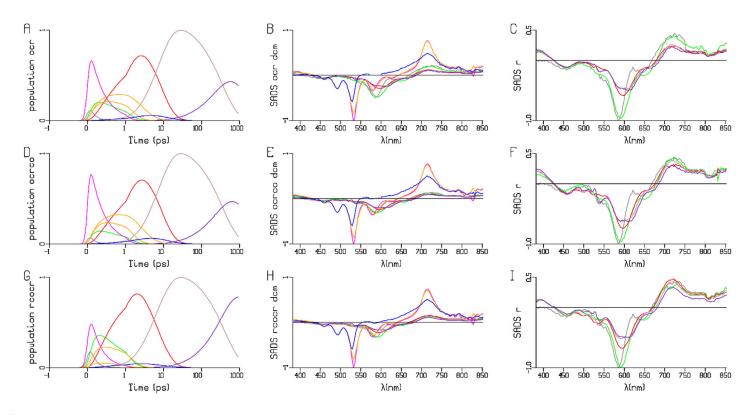


Figure S 22. Populations of the target kinetic schemes from Figure S 21 (A,D,G), estimated SADS (B,E,H) and SADS of the five \mathbf{r}^* states (C,F,I) of **ocr**, **rcocr** and **ocrco** in CH₂Cl₂. Key: magenta, **o*****FC**: excited **o** chromophore in the Franck-Condon state; orange, two different conformations of **o*** with identical SADS; blue, **ocRP**: $\mathbf{c}^+\mathbf{o}^-$ radical ion pair; grey, green, red, brown, purple: successively relaxed \mathbf{r}_1^* , \mathbf{r}_2^* , \mathbf{r}_3^* , \mathbf{r}_4^* , \mathbf{r}_5^* .

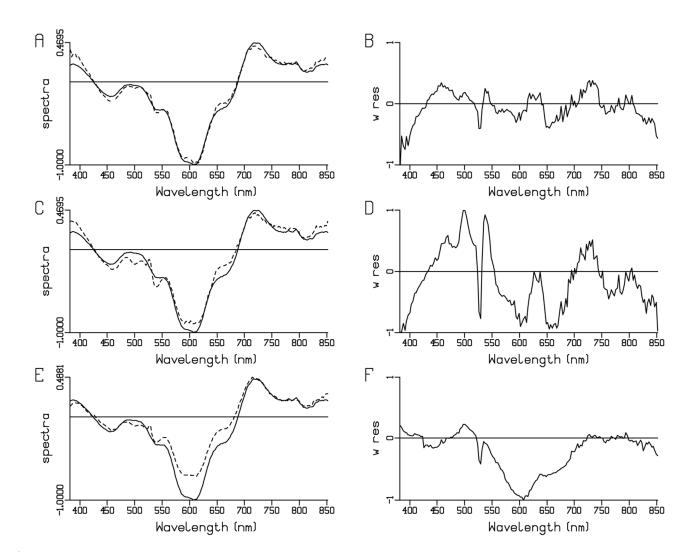


Figure S 23. Overlay of the final **r*** SADS from **rc** (solid lines in panels A,C,E) and from **ocr**, **ocrco** and **rcocr** in CH2Cl2 (dashed lines in panels A,C,E, respectively). The difference between the final **r*** SADS from **rc** and from **ocr**, **ocrco** and **rcocr** is depicted in panels B,D,F, respectively. Note the **o** bleach features near 490 and 530 nm.

Spectral evolution in ocrcg and gcrcocrcg

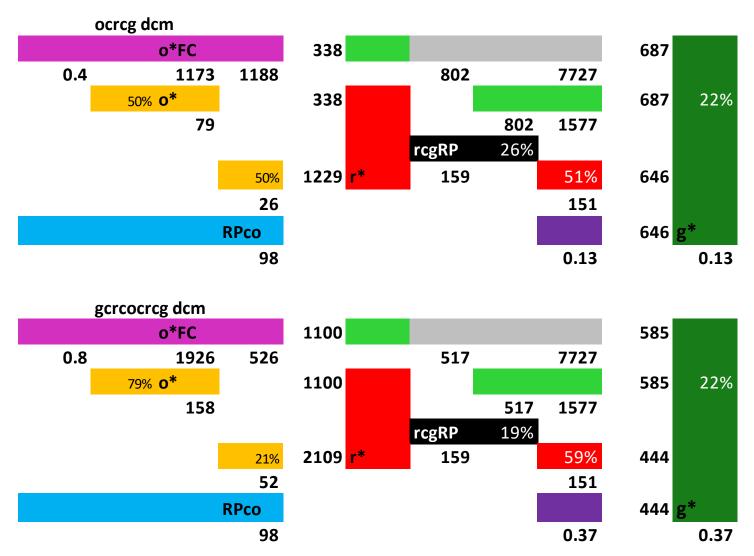


Figure S 24. Kinetic schemes used for the simultaneous target analysis of **ocrcg** and **gcrcocrcg** in CH₂Cl₂. All rate constants in ns⁻¹. Key: magenta, **o*FC**: excited **o** chromophore in the Franck-Condon state; orange, two different conformations of **o*** with identical SADS; blue, **ocRP**: $c^+ \cdot o^-$ radical ion pair; grey, green, red, purple: successively relaxed **r**₁*, **r**₂*, **r**₃*, **r**₄*; dark green: **g***; black, **rcgRP**: **rcg** radical pair.

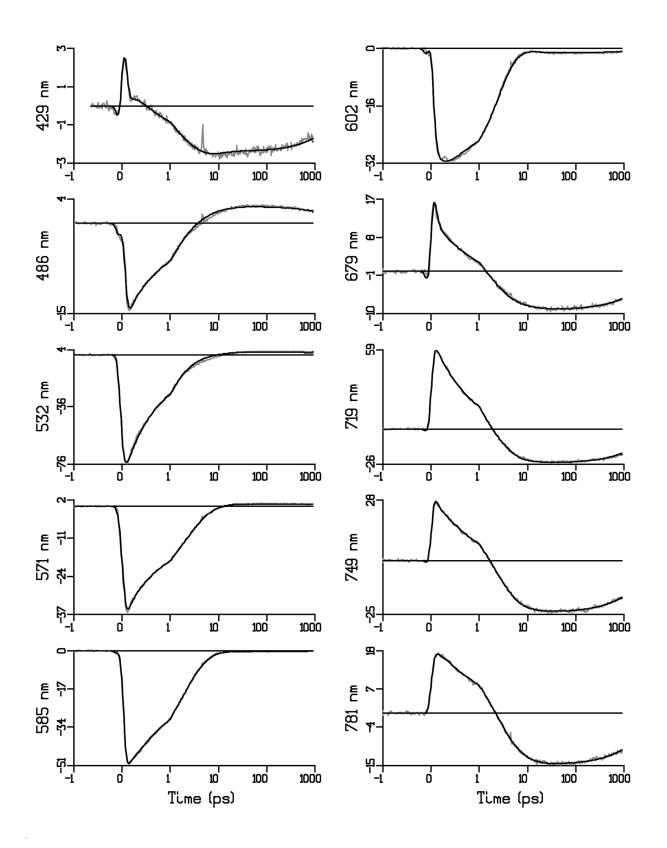


Figure S 25. Selected time traces of **gcrcocrcg** in CH₂Cl₂ after excitation at 530 nm data (in mOD, grey) and fit (black). Wavelength is indicated in the ordinate label. Note that the time axis is linear until 1 ps (after the maximum of the IRF), and logarithmic thereafter.

References

[37] C. Hippius, Multichromophoric Arrays of Perylene Bisimide Dyes - Synthesis and Optical Properties; Multichromophore Perylenbisimidkaskaden - Synthese und optische Eigenschaften, in, Universität Würzburg, Fakultät für Chemie und Pharmazie, Würzburg, 2007, pp. 237.