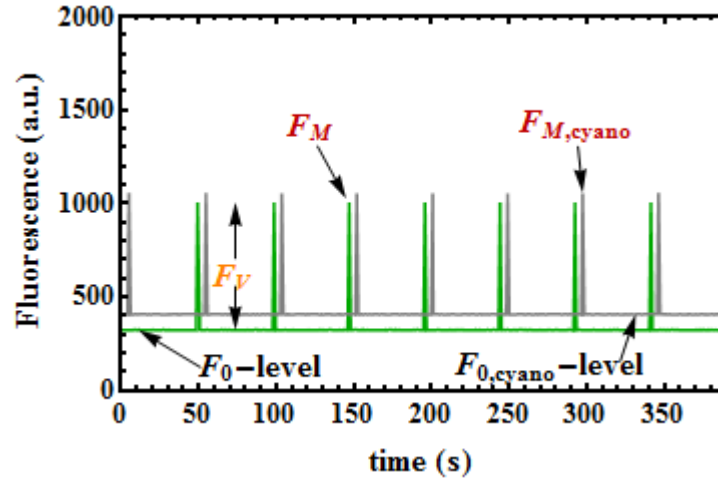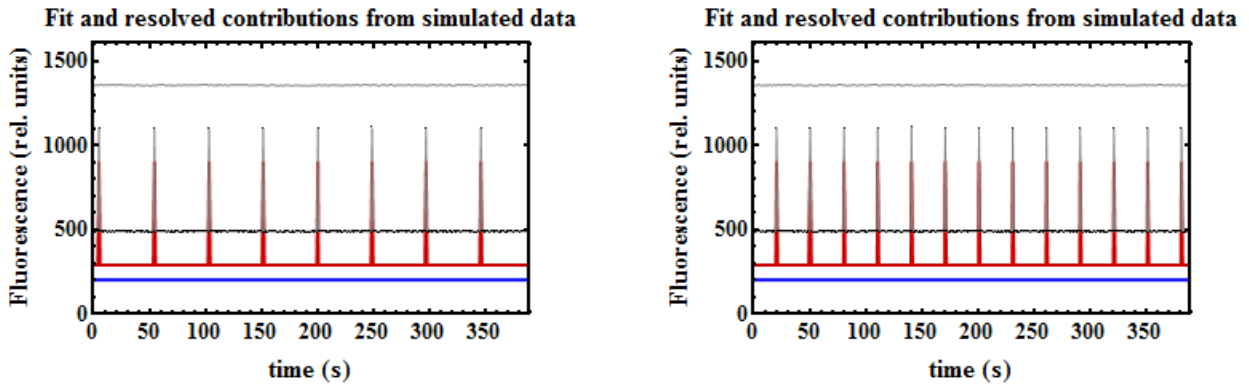# Supporting Information

Figure S 1 Simulated PAM–signal of *i)* dark–adapted PSII as it is expected in plants (green) and *ii)* PSII and with an extra phycobilin contribution as it is expected in cyanobacteria (gray). A saturation pulse has been applied every 50 s (dephased for visibility).

Figure S 2 Simulated data with noise level 0.01 and a pulse every 50 s (left panel) and 20 s (right panel). These datasets have been fitted with the model containing two components: PPc2 and PB_{free}. Key: Black dots: simulated data points; Red: PPc2-contribution; Blue: PB_{free}-contribution; Gray: sum of the two contributions. Residuals are shown on top with an offset for better visibility.

On the relationship between Chl a fluorescence and photochemistry.

Three essential assumptions are made in order to use fluorescence as a proxy for photosynthetic activity. First, harvested light has three possible pathways: either it reaches the reaction centers of PSII where it is used for photochemistry (Φ_{ph}) or it gives rise to fluorescence (Φ_{fl}) or it gets dissipated as heat (Φ_{h}):

$$\Phi_{ph} + \Phi_{fl} + \Phi_{h} = 1 \tag{S1}$$

Second, a saturation–pulse induces a state where Q_A is reduced, *i.e.* transient saturation of PSII (closed centers), while the fluorescence and heat dissipation yields are maximal:

$$(\Phi_{fl})_M + (\Phi_{h})_M = 1 \tag{S2}$$

And last, the ratio of the fluorescence and heat dissipation yields is assumed not to vary during the saturating pulse:

$$\frac{\Phi_{h}}{\Phi_{fl}} = \frac{(\Phi_{h})_M}{(\Phi_{fl})_M} \tag{S3}$$

Therefore, combining Eq. (S2) and (S3) we obtain an expression for $(\Phi_h)_M$ that can be replaced in Eq. (S1) so that the yield of photochemistry is strictly related to the one of fluorescence:

$$\Phi_{ph} = 1 - \Phi_{fl} - \left(\frac{\Phi_{fl}}{(\Phi_{fl})_M} - \Phi_{fl} \right) = \frac{(\Phi_{fl})_M - \Phi_{fl}}{(\Phi_{fl})_M} \tag{S4}$$

This relationship has propelled numerous studies aiming to track fluorescence under controlled light conditions and relate those measurements to the PSII photochemical yield. The parameter:

$$\eta = \frac{F_V}{F_M} = \frac{F_M - F_0}{F_M} \tag{S5}$$

directly derived from Eq.(S4), has been commonly equaled to the index of maximal photochemical efficiency of PSII in plant studies. We therefore obtain (see Eqs. (1)& (2)):

$$\eta_{cyano} = \frac{F_{M,cyano} - F_{0,cyano}}{F_{M,cyano}} = \frac{F_{M,PSII} - F_{0,PSII}}{F_{M,PSII} + F_{0,PB}} \tag{S6}$$

Hence, for a fixed level of Chl *a* (PSII) fluorescence, the cyanobacterial η_{cyano} decreases with increasing phycobilisome/Chl ratio or disconnected PBs/PSII (for an example under Fe starvation see Wilson et al. 2007). Figure S 3 illustrates Eq. (S6). Note that this decrease is *not* related to a change in photochemistry. Thus, without knowing how much the PB contributes to the signal, the parameter F_V/F_M cannot be used to measure the maximal photochemical efficiency of PSII. In addition, when performed

in plants, these measurements are done in darkness, conditions in which there is no NPQ. In contrast, cyanobacterial cells tend to state 2 in darkness Campbell et al. 1998; as a consequence F_M is quenched.
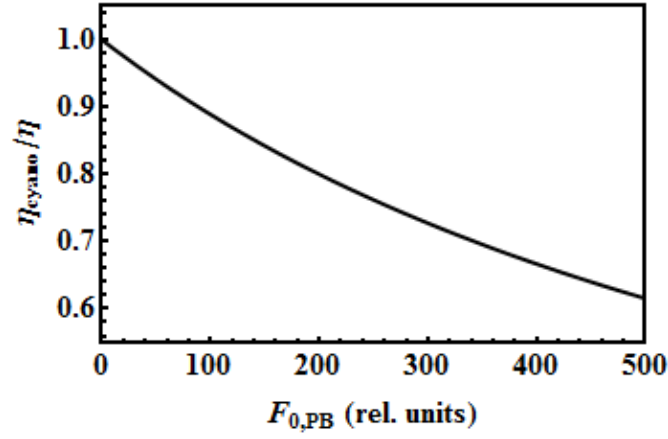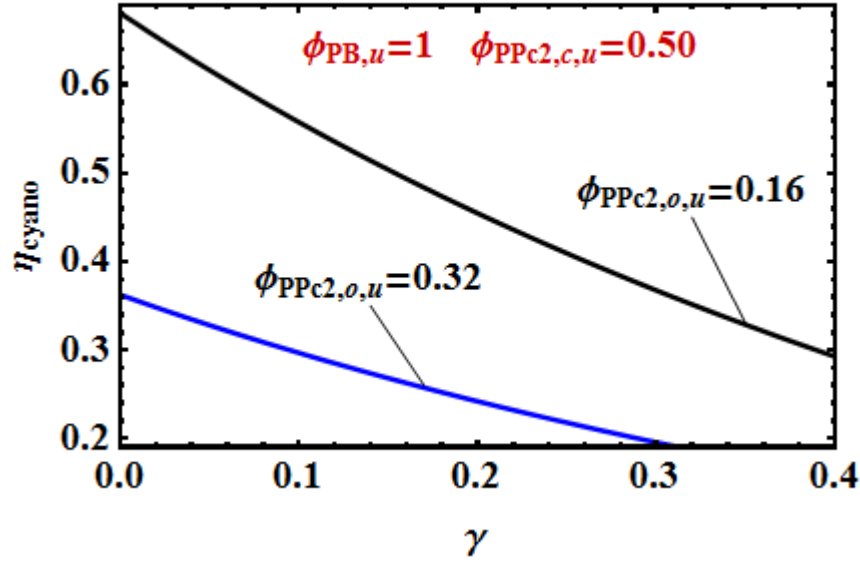

Figure S 3 Normalized expression for cyanobacterial η_{cyano} (with respect to plants) as a function of the phycobilin-related contribution $F_{0,PB}$.

Using Eq. (3) we can re-write Eq.(S6):

$$\eta_{cyano} = \frac{c_2 \square (\phi_{PPc2,c,u} - \phi_{PPc2,o,u})}{c_2 \square \phi_{PPc2,c,u} + \gamma \square \phi_{PB,u}} \tag{S7}$$

with:

$$c_2 + \gamma = 1 \tag{S8}$$

Figure S 4 η_{cyano} as a function of the relative concentration of PB_{free} and the difference of PPc2 quantum yields in open and closed states. Two curves have been plotted for two distinct values of $\phi_{PPc2,o,u}$ while $\phi_{PPc2,c,u}$ has been kept constant. The graph illustrates how the variable fluorescence decreases with decreasing contrast between open and closed states: $\Delta\phi_{PSII} = \phi_{PPc2,c,u} - \phi_{PPc2,o,u}$. Key: Black: $\Delta\phi_{PSII} = 0.34$; Blue: $\Delta\phi_{PSII} = 0.18$.

Derivation of the quenching and recovery dynamics.
Formation of a quenching complex C_q. The homogeneous system. The system of second order differential equations describing the formation of a quenching complex C_q from an activated OCP^r that binds to a PB is:

$$PB'(t) = -k_1 \cdot PB(t) \cdot OCP^r(t) + k_2 \cdot C_q(t) \tag{S9}$$

$$OCP^{r'}(t) = -k_1 \cdot PB(t) \cdot OCP^r(t) + k_2 \cdot C_q(t) \tag{S10}$$

$$C_q'(t) = k_1 \cdot PB(t) \cdot OCP^r(t) - k_2 \cdot C_q(t) \tag{S11}$$

where the product $PB(t) \cdot OCP^r(t)$ accounts for the encounter probability between a photoactivated OCP and a PB.

Note that $PB'(t) = OCP^{r'}(t) = -C_q'(t)$. This property allows the system to be simplified and expressed in terms of the single variable $PB(t)$.

After subtraction of Eq. (S9) from Eq. (S10) we obtain the homogeneous equation:

4

$$\int \left(OCP^{r\prime}(t) - PB'(t) \right) dt = 0 \qquad (S12)$$

from which it follows that the general solution is:

$$OCP^{r}(t) - PB(t) = a_1 \qquad (S13)$$

where a_1 is an arbitrary constant. We obtain the value of a_1 by simply evaluating both functions at time t=0. Thus:

$$a_1 = OCP^{r}{}_0 - PB_0 \qquad (S14)$$

Replacing Eq. (S14) in Eq. (S13) yields an expression of $OCP^{r}(t)$ as a function of PB(t) given some starting values for the PB and photoactivated form of OCP at time t=0:

$$OCP^{r}(t) = PB(t) - PB_0 + OCP^{r}{}_0 \qquad (S15)$$

Furthermore, adding Eq. (S9) to Eq. (S10) leads to a row of equations similar to Eqs.(S11)–(S15) in the case of the quenching complex $Cq(t)$. We obtain the final expression:
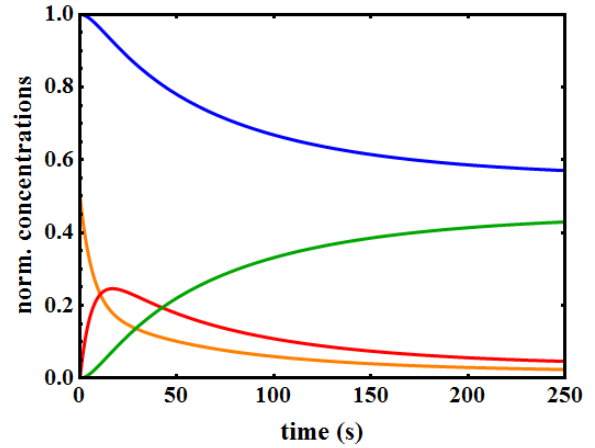
$$C_q(t) = -PB(t) + PB_0 + C_{q,0} \qquad (S16)$$

By inserting Eqs. (S15) & (S16) in Eq. (S11) we finally obtain a single equation of second order for the single variable PB(t). Its graphical solution is shown in Figure S 5A :

$$PB'(t) = -k_1 \cdot PB^2(t) + \left(k_1 \cdot (PB_0 - OCP^{r}{}_0) - k_2 \right) \cdot PB(t) + k_2 \cdot (PB_0 + C_{q,0}) \qquad (S17)$$

5

A: Parameters:
k_1=0.025; k_2=0.001; OCP^r_0=0.5; OCP^o_0=0
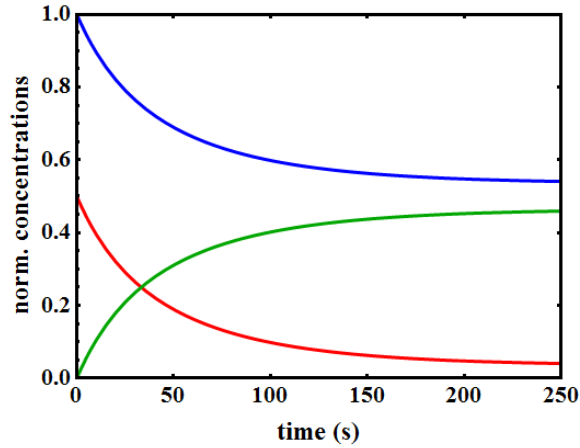
B: Parameters:
k_1=0.025; k_2=0.001; OCP^r_0=0; OCP^o_0=0.5

C: Parameters:
k_1=0.025; k_2=0.01; OCP^r_0=0; OCP^o_0=0.5
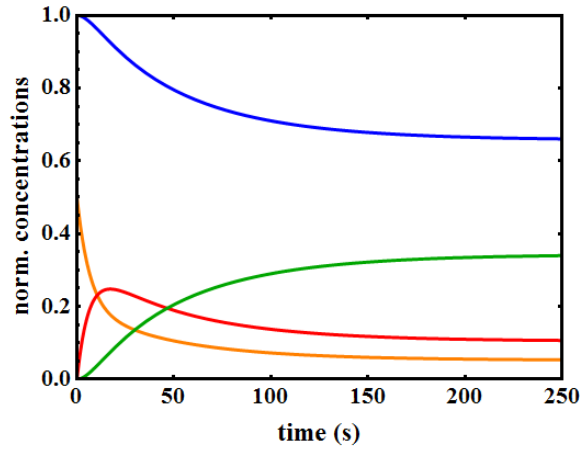
D: Parameters:
k_1=0.025; k_2=0.001; OCP^r_0=0; OCP^o_0=0.33
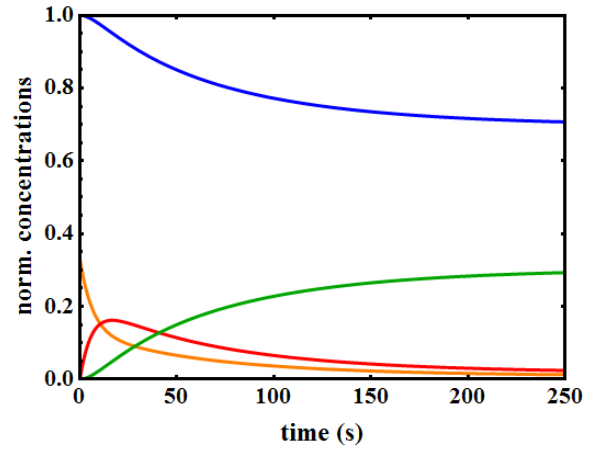
Figure S 5 Graphical solutions shown for a homogeneous (A) and an inhomogeneous (B–D) system of differential equations reproducing the NPQ-dynamics over time. In all panels: $[PB_0]$=1, $[C_q]$=0. In panels B–D: κ_1=0.1, k_3=0.05. Key: blue: Q_u; red: Q_r; green: Q_q; orange: Q_o (see text for explanation).

The light-intensity dependent generation of OCP^r. The inhomogeneous system. The observed kinetics up to now imply a non–zero concentration of OCP^r at time t=0 (note $OCP^r{}_0$=0.5 in Figure S 5A). This is, however, far from the experimental observation: OCP has to be photo–converted by light absorption from its orange to its red form first. This process depends on the wavelength and intensity of the light.

The time–dependent generation of OCP^r from the orange form of a fixed amount of OCP^o will therefore be introduced as an inhomogeneity in Eq. (S12) with κ_I being the light-intensity dependent formation rate of OCP^r:

$$\int \left(OCP^{r\prime}(t) - PB'(t) \right) dt = \int \kappa_I \cdot OCP^o{}_0 \cdot e^{-\kappa_I \cdot t} \, dt \tag{S18}$$

which once integrated and evaluated at time t=0 (for the left side of the equation we know the answer already from Eq. (S13)):

$$a_1 = -OCP^o{}_0 \cdot e^{-\kappa_I \cdot t} \Big|_{t=0} + a_2 \tag{S19}$$

yields the constant $a_2 = a_1 + OCP^o{}_0$ and we obtain thus the solution for all times t:

$$OCP^r(t) = PB(t) - PB_0 + OCP^r{}_0 + OCP^o{}_0 \cdot \left(1 - e^{-\kappa_I \cdot t} \right) \tag{S20}$$

An analogous equation to Eq. (S17) can now therefore be derived by inserting Eq. (S20) instead of Eq. (S15) in Eq. (S9) (see Eq. (S21)).

Finally, we introduce a last kinetic parameter, namely a deactivation rate of OCP^r k_3 that accounts for the fact that the population OCP^o grows as OCP^r deactivates. The system of differential equations we need to solve reads:

$$PB'(t) = -k_1 \cdot PB^2(t) + \left(k_1 \cdot \left(PB_0 - OCP^r{}_0 - OCP^o{}_0 + OCP^o(t) \right) - k_2 \right) \cdot PB(t) + k_2 \cdot (C_{q,0} + PB_0) \tag{S21}$$

$$OCP^{o\prime}(t) = -\kappa_I \cdot OCP^o(t) + k_3 \cdot \left(PB(t) - PB_0 + OCP^r{}_0 + OCP^o{}_0 - OCP^o(t) \right) \tag{S22}$$

A solution to this system can only be given numerically. The graphical solution of this quenching function which we call $Q_i(\boldsymbol{P},t)$ is shown in Figure S 5 for a chosen set $\boldsymbol{P}$ of parameters. $Q_i(\boldsymbol{P},t)$ has four components: Q_u (blue) describes de unquenched PB, Q_q (green) the quenching complex C_q, Q_o (orange) the orange form of OCP and Q_r its photo–converted red form. Notice that this time $OCP^r{}_0$=0 while $OCP^o{}_0$=0.5. The deactivation rate of OCP^r k_3 has been chosen to be smaller (k_3=0.05) than the activation rate (κ_I=0.1).

The recovery phase. Turning off the actinic light simply translates in setting the parameter κ_I to zero. Furthermore, we assume that during the quenching phase only a fraction of the orange OCP is converted to its red form. The graphical solution of the recovery phase with two different sets of parameters (k_2 is varied) is shown in Figure S 6.

A: Parameters:

$\kappa_I=0$; $k_2=0.001$

B: Parameters:
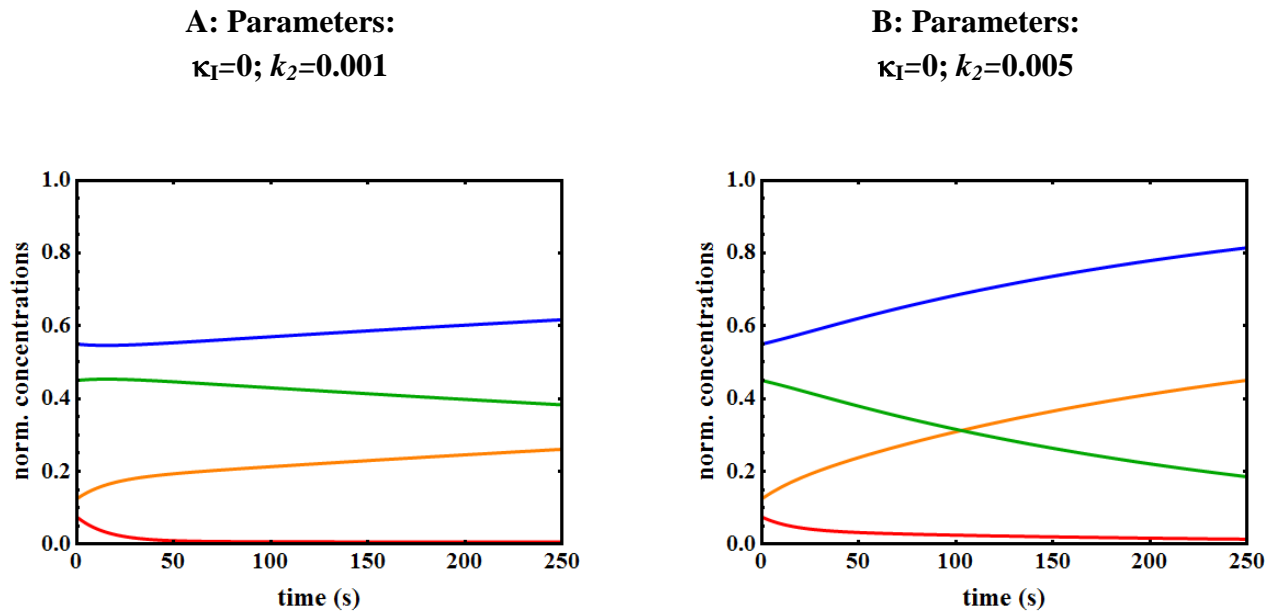
$\kappa_I=0$; $k_2=0.005$

Figure S 6 Graphical solution for the recovery dynamics ($\kappa_I=0$) with $k_2=0.001$ (A) and $k_2=0.005$ (B). Parameters: $PB_0=0.55$; $C_{q,0}=0.45$; $OCP^r_0=0.1$; $OCP^o_0=0.001$; $k_1=0.025$; $k_3=0.05$. Key: blue: Q_u; red: Q_r; green: Q_q; orange: Q_o

NPQ Model applications

Light-dependent $OCP^o \rightarrow OCP^r$ conversion.

The NPQ dynamics in the cyanobacterium *Synechocystis* has been studied by Gorbunov et al. 2011. In Fig. 3B of their publication they show how NPQ is differently induced when varying the ambient light intensity. We have digitized their data using Plot Digitizer, an open source tool to create readable files from graphs. The digitized data has then been fitted with our model (see left panel of Figure S 7). The light–intensity dependent rate is κ_I, i.e. the rate with which the OCP intermediate state is formed in the Gorbunov's model. Their result shows that the rate of NPQ induction accelerates with an increase in photon flux density of actinic blue light Gorbunov et al. 2011. We have modelled the same effect by choosing a set of realistic starting values. Two parameters have been estimated: the $OCP^o \rightarrow OCP^r$ conversion rate κ_I and the initial concentration of OCP in its orange form, [OCP^o].

Table S 1 shows the results. The initial concentration [OCP^o] is in the vicinity of 60% ([PB]=100%) for all the samples. The behaviour of κ_I as a function of the light intensity is the one expected: the more intense the ambient actinic light, the quicker the photoactivation of OCP, and therefore, the quenching process. A scatter plot relates the estimated κ_I to the light intensity value (right panel in Figure S 7). In

principle, the model can be used to find adequate κ_I values given an unknown ambient actinic light intensity via a simple linear regression.
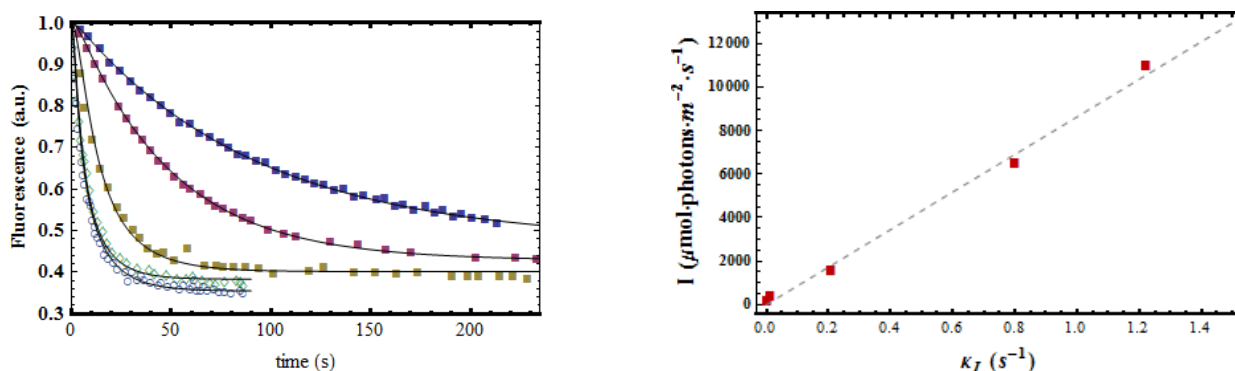
Figure S 7 *Left:* Light intensity dependence of NPQ induction as studied by Gorbunov, *et al.* Key: 200 (blue closed squares), 400 (magenta closed squares), 1600 (yellow closed squares), 6500 (green open diamonds) and 11000 (blue open circles) $\mu mol \cdot m^{-2} \cdot s^{-1}$. Fit curves are shown in black. *Right:* Scatter plot displaying the conversion rates κ_I as estimated from the fits against ambient actinic light's intensity as given by Gorbunov, *et al.* A linear regression is also shown in gray.

I (μE)	200	400	1600	6500	11000
[OCP^o]	0.54	0.58	0.64	0.64	0.68
κ_I	0.01	0.02	0.22	0.81	1.23

Table S 1 Estimated parameters applying our model to the digitized data of Gorbunov, *et al.*

Estimating the OCP/PB ratio of in vitro experiments.

Our model was also able to reproduce fluorescence decrease as first observed *in vitro* by Gwizdala et al. 2011 and published in Jallet et al. 2012. NPQ was induced by blue–green light of 900 $\mu mol \cdot m^{-2} \cdot s^{-1}$ (left panel in Figure S 8) in samples containing a phosphate concentration of 0.5 M, where the ratio OCP/PB was varied (4, 8, 20 and 40). Since these experiments were carried out *in vitro* some of the kinetic parameters differ slightly from the Gorbunov example discussed above. The data was satisfactorily fitted and the different ratios OCP/PB (shown next to the curves in Figure S 8) were estimated. They agree surprisingly well with the experimental conditions chosen by Jallet, *et al.*, the greatest error being of ca. 10% in the data point OCP/PB$^{(exp)}$ = 40 (OCP/PB$^{(est)}$ = 51).

Jallet, *et al.* carried out an additional experiment with the phosphate concentration being 0.8 M. Even though the model does not include any *buffer parameter* and it is far from predicting any conformational changes, we assumed that these effects would ultimately translate in affected OCP binding/detaching rates. Thus, we were still able to fit the data (Figure S 9) and extract again the ratios OCP/PB$^{(est)}$. The results agreed fairly well with the experimental results.
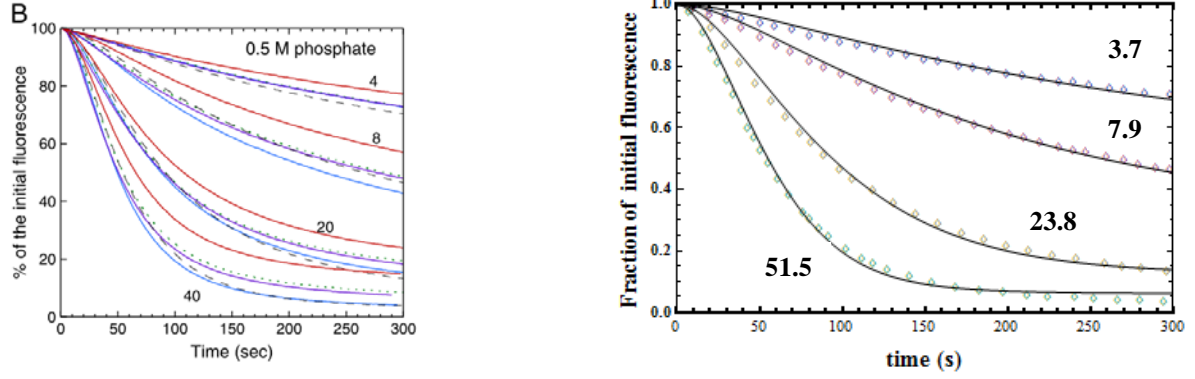
Figure S 8 *Left:* Fluorescence decrease induced with blue-green light (900 μmol·m^{-2}·s^{-1}) with different OCP to PBs ratios (4, 8, 20 and 40) in samples containing 0.5 M Phosphate as published by Jallet, et al. (2012). *Right:* Digitized wild type data (black dashed curves on the left panel) fitted using our NPQ model. The only parameter varying from curve to curve is the initial concentration [OCP^o] which results in different OCP to PBs ratios. The estimated ratios are shown next to the corresponding fit.
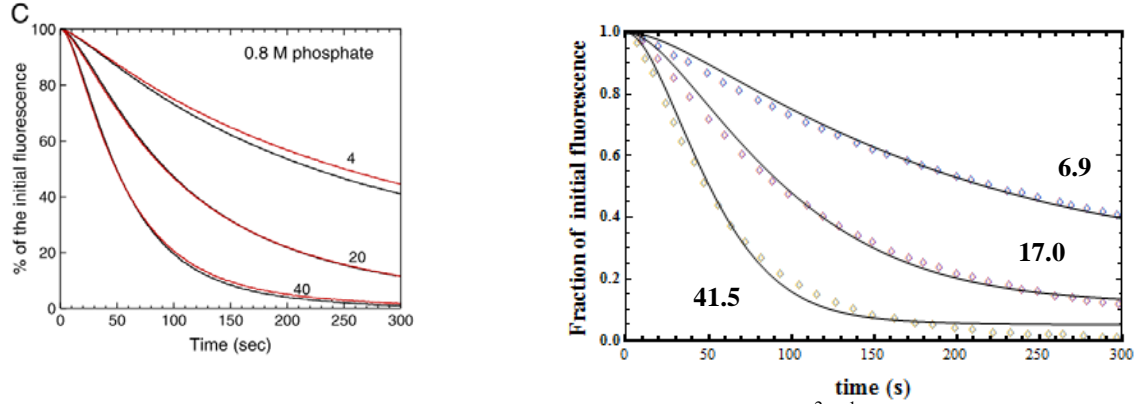
Figure S 9 *Left:* Fluorescence decrease induced with blue-green light (900 μmol·m^{-2}·s^{-1}) with different OCP to PBs ratios (4, 20 and 40) in samples containing 0.8 M Phosphate as published by Jallet, et al. (2012). *Right:* Digitized data fitted using our NPQ model. The only parameter varying from curve to curve is the initial concentration [OCP^o] which results in different OCP to PBs ratios. The estimated ratios are shown next to the corresponding fit.

The PSII$_{free}$ contribution.

Alternatively, we can write Eq. (6) as:

$$J_{PAM} = \gamma \cdot \Phi_{PB,u} + c_2 \cdot \Phi_{PPc2,o,u} + f_2 \cdot \Phi_{PSII,o} + \Delta(t) \qquad (S23)$$
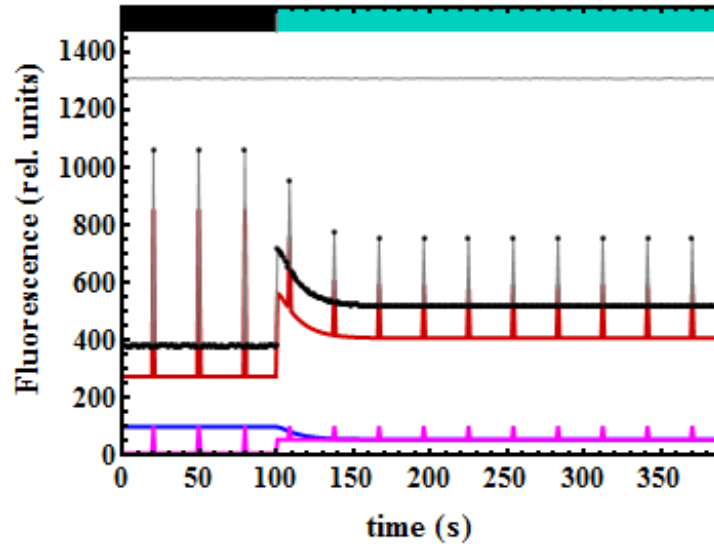
with:

$$\Delta(t) = \sum_{i}^{n_{pulse}} \left(c_2(\Phi_{PPc2,c,u} - \Phi_{PPc2,o,u}) + f_2(\Phi_{PSII,c} - \Phi_{PSII,o}) \right) \cdot \delta(t - t_i) \qquad (S24)$$

Eq. (S24) helps realizing one important feature: the F_M–level (and therefore the variable fluorescence) is determined by the sum of the differences in quantum yields between the species in closed and open states. Eq. (S6) with an additional PSII$_{free}$ component (f_2):

$$\eta_{cyano} = \frac{c_2 \square (\phi_{PPc2,c,u} - \phi_{PPc2,o,u}) + f_2 \square (\phi_{PSII,c} - \phi_{PSII,o})}{c_2 \square \phi_{PPc2,c,u} + f_2 \square \phi_{PSII,c} + \gamma \square \phi_{PB,u}} \qquad (S25)$$

$$c_2 + f_2 + \gamma = 1 \qquad (S26)$$

Figure S 10 Simulated data with a pulse every 30 s and noise levels of 0.01. At time t=100 s strong blue-green light is turned on. Its intensity is such that a fraction c_0 of the RCs gets closed and the OCP^o→OCP^r conversion takes place with κ_I=0.09s^{-1}. The amount of OCP^r formed is [OCP^r]=0.5 and it binds to PB with k_I=0.30 s^{-1}. Parameters have been estimated from this data in trial Q1. Key: Black dots: simulated data points; Red: PPc2-contribution; Blue: PB$_{free}$-contribution; Magenta: PSII$_{free}$-contribution; Gray: sum of the three contributions. Residuals are shown on top with an offset of 1300.

11

Solving the linear system for three parameters

The linear system formulated in Eq. (11):

$$\begin{pmatrix} F_{0,cyano} \\ F_{M,cyano} \\ F^{q'}_{M,cyano} \end{pmatrix} = \begin{pmatrix} \phi_{PB,u} & \phi_{PPc2,o,u} & \phi_{PSII,o} \\ \phi_{PB,u} & \phi_{PPc2,c,u} & \phi_{PSII,c} \\ \phi_{PB,q} & \phi_{PPc2,c,q} & \phi_{PSII,c} \end{pmatrix} \square \begin{pmatrix} \gamma \\ c_2 \\ f_2 \end{pmatrix}$$

has a unique solution if the **A** matrix can be brought to triangular form. We study the **A** matrix by means of a simple Gaussian elimination calculation to determine under which conditions the system has a solution. Elimination of the off-diagonal elements of the first column yields:

$$\begin{matrix} \phi_{PB,u} & \phi_{PPc2,o,u} & \phi_{PSII,o} \\ 0 & \phi_{PPc2,c,u} - \phi_{PPc2,o,u} & \phi_{PSII,c} - \phi_{PSII,o} \\ 0 & \phi_{PB,u} \cdot \phi_{PPc2,c,q} - \phi_{PB,q} \cdot \phi_{PPc2,o,u} & \phi_{PB,u} \cdot \phi_{PSII,c} - \phi_{PB,q} \cdot \phi_{PSII,o} \end{matrix}$$

In as second step, we would need to eliminate the third–row element of the second column to get a triangular matrix. Alternatively, we could first exchange columns 2 and 3 and go on with the elimination step by multiplying a factor that cancels out the diagonal element of the second column. The first choice yields:

$$\begin{matrix} \phi_{PB,u} & \phi_{PPc2,o,u} & \phi_{PSII,o} & \\ 0 & \phi_{PPc2,c,u} - \phi_{PPc2,o,u} & \phi_{PSII,c} - \phi_{PSII,o} & \Big| \dfrac{\phi_{PB,u} \cdot \phi_{PPc2,c,q} - \phi_{PB,q} \cdot \phi_{PPc2,o,u}}{\phi_{PPc2,c,u} - \phi_{PPc2,o,u}} \\ 0 & \phi_{PB,u} \cdot \phi_{PPc2,c,q} - \phi_{PB,q} \cdot \phi_{PPc2,o,u} & \phi_{PB,u} \cdot \phi_{PSII,c} - \phi_{PB,q} \cdot \phi_{PSII,o} & \end{matrix}$$

Notice, however, that either way we choose to proceed, a multiplying factor is needed in which the denominator is the difference in fluorescence quantum yields of the closed and open states of either PPc2 or PSII_{free}. Hence, this simple calculation shows that, unless there is neat contrast between open and closed states:

$$\phi_{PPc2,c,u} - \phi_{PPc2,o,u} \neq 0 \tag{S27}$$

$$\phi_{PSII,c} - \phi_{PSII,o} \neq 0 \tag{S28}$$

the **A** matrix cannot be brought to a triangular form and the system cannot be solved. This result was already qualitatively contained in Eq.(S24).
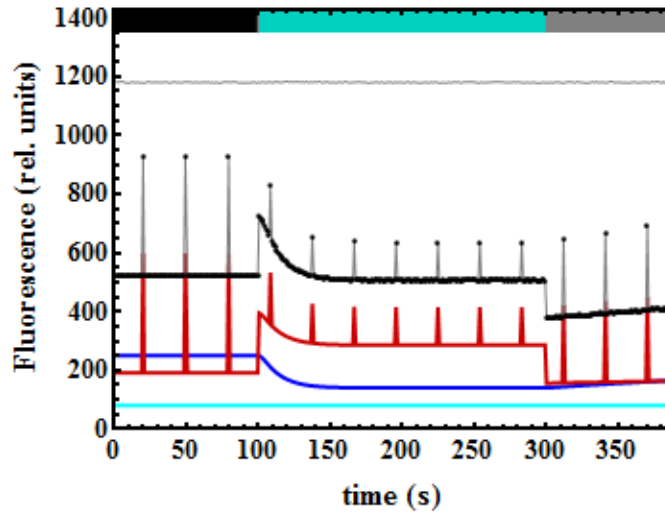
The PSI contribution.

$$F_{0,cyano} = \gamma\Box\phi_{PB,u} + c_2\Box\phi_{PPc2,o,u} + \beta\Box c_2\Box\phi_{PSI} \tag{S29}$$

$$F_{M,cyano} = \gamma\Box\phi_{PB,u} + c_2\Box\phi_{PPc2,c,u} + \beta\Box c_2\Box\phi_{PSI} \tag{S30}$$

$$F^{q'}_{0,cyano} = \gamma\Box\phi_{PB,q} + (1-c_0)\Box\left(c_2\Box\phi_{PPc2,o,q}\right) + c_0\Box\left(c_2\Box\phi_{PPc2,c,q}\right) + \beta\Box c_2\Box\phi_{PSI} \tag{S31}$$

$$F^{q'}_{M,cyano} = \gamma\Box\phi_{PB,q} + c_2\Box\phi_{PPc2,c,q} + \beta\Box c_2\Box\phi_{PSI} \tag{S32}$$

Figure S 11 Same caption as in Figure S 10. A PSI_{free}-contribution has been added instead of $PSII_{free}$. The relative contributions used for simulation were: γ=0.05, β=3, *scale*=5000. Estimated parameters: (γ=0.0498 +/- 0.0003), (β=3.01 +/- 0.02), (*scale*=5008.5 +/- 18.6). Key: Black dots: simulated data points; Red: PPc2-contribution; Blue: PB_{free}-contribution; Cyan: PSI_{free}-contribution; Gray: sum of the three contributions. Residuals are shown on top with an offset of 1300.

Note that in the simulation of Figure S 11 all NPQ parameters had to be kept fixed. When freeing those parameters, it was found that β correlates strongly with the *scale* and NPQ parameters and cannot be accurately estimated due to numerical unidentifiability.

List of parameters

γ – concentration of functionally uncoupled PB
c_2 – concentration of PB–PSII–complexes
f_1 – concentration of free PSI
f_2 – concentration of free PSII
β – proportionality factor between the number of PSI–trimers and PSII–dimers
c_0 – fraction of closed photosystems when turning the strong blue–green light on
k_1 – OCP to PB binding rate
k_2 – FRP detaching rate of OCP
k_3 – OCP^r deactivation rate
κ_I – OCP^o to OCP^r activation rate
OCP$^o_0$ – initial concentration of OCP^o
scale – scaling factor matching the actual fluorescence scale of the fluorometer

parameter	trial Q1 estimate	st. error	t-val	trial Q2 estimate	st. error	t-val	trial Q3 estimate	st. error	t-val	trial Q4 estimate	st. error	t-val	trial Q+R estimate	st. error	t-val
γ	0.0509	0.0011	92.5	0.0496	0.0006	88.9	0.0505	0.0010	48.6	0.0505	0.0010	48.6	0.0504	0.0004	119.9
f_2	0.1042	0.0041	49.5	0.0987	0.0021	47.8	0.1032	0.0049	21.2	0.1028	0.0049	21.2	0.1024	0.0015	68.8
c_0	0.4998	0.0008	1370.3	0.4997	0.0004	1351.9	0.5003	0.0007	687.5	0.5002	0.0007	682.8	0.4999	0.0005	1059.7
$scale$	1997.67	4.39	948.6	2001.79	2.21	904.6	1999.83	2.94	680.7	1999.64	2.94	680.4	1999.53	1.81	1102.4
k_1	0.3	-	-	0.3002	0.0029	103.2	0.3043	0.0051	60.0	0.287	0.014	20.5	0.2990	0.0033	90.1
k_2	0.003	-	-	0.003	-	-	0.0025	0.0005	4.8	0.0067	0.0034	2.0	0.00306	0.00004	83.2
OCP^o_0	0.5	-	-	0.5	-	-	0.5	-	-	0.514	0.012	43.5	0.5012	0.0007	664.7

Table S 2 Parameter estimation of NPQ parameters in several trials: from trial Q1-4 the kinetic parameters have been freed stepwise. The t-values, in particular that of k_2, gradually drop. Adding a region of fluorescence recovery (trial Q+R) allows all parameters to be estimated with acceptable t-statistics. The Q+R trials were performed for the case of $PSII_{free}$ being added and the amount of parameters being increased gradually. t-val: ratio of estimate and its standard error.

parameter	WT rmse = 30.9 estimate	st. error	t-val	ΔApcD rmse = 31.3 estimate	st. error	t-val	ΔApcF rmse = 27.2 estimate	st. error	t-val	ΔApcDF rmse = 27.2 estimate	st. error	t-val
γ	0.034	0.001	28.0	0.059	0.002	36.0	0.072	0.002	46.1	0.131	0.002	56.8
β (fixed)	3.6			3.6			3.6			3.6		
c_0	0.606	0.011	54.8	0.564	0.011	53.1	0.571	0.011	50.7	0.577	0.013	46.0
$scale$	13899.6	181.2	76.7	14314.8	193.4	74.0	14838.5	177.1	83.8	13942.9	169.0	82.5
k_1	0.121	0.010	11.7	0.243	0.018	13.7	0.302	0.031	9.7	0.451	0.049	9.3
k_2	0.0096	0.0004	21.7	0.0068	0.0002	29.7	0.019	0.001	18.7	0.018	0.001	24.2
OCP^o_0	0.61	0.01	48.9	0.64	0.007	88.9	0.53	0.01	61.2	0.46	0.01	91.8

Table S 3 Parameter estimation of experimental data. PAM was performed on whole cells of WT *Synechocystis* and mutants thereof. Estimates with their standard error and associated *t*-values. Rmse: root mean square error of the fit, t-val: ratio of estimate and its standard error.