

Gas-Phase Electronic Absorption Spectra of Deuterated Linear Seven-Carbon Radicals

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Rotationally resolved gas-phase absorption spectra of partially and fully deuterated linear seven-carbon chain radicals are presented in this article. The carbon-based molecules are generated in a supersonically expanding planar plasma by discharging a gas mixture of acetylene and deuterium-enriched acetylene in helium and argon. Spectra are recorded in direct absorption using cavity ring-down spectroscopy. The rotational analyses of the present experimental spectra allow to determine both ground and excited state rotational constants, as well as the upper state band origins of the two deuterated species.

Keywords: carbon chain radicals, cavity ring-down spectroscopy, electronic spectroscopy, HC₇D and DC₇D

Introduction

Carbon is a major player in the chemistry of the dilute gas in clouds between the stars, because of its abundance and its ability to form complex species. While 75% of the molecules detected in interstellar space are carbon-bearing, they are also key elements in the evolution of prebiotic molecules.^[1] Highly unsaturated carbon chain radicals have been identified in dark interstellar clouds following radio-astronomical observations.^[2,3] Centro-symmetric chains such as NC_nN⁽⁺⁾ and $HC_nH^{(+)}$ are expected to exist in the interstellar medium as well but are radio-silent given their lack of a permanent dipole moment.^[4] These chains are only detectable via lowlying bending modes, typically in the sub-millimeter region, or via infrared spectra of vibrational bands.^[5-7] Optical transitions offer another alternative, particularly towards diffuse and translucent clouds, where electronic transitions of carbon chains have been associated with diffuse interstellar band features.^[8]

The electronic $\widetilde{A}^{3}\Sigma_{u}^{-} - \widetilde{X}^{3}\Sigma_{g}^{-}$ transitions of the oddpolypne HC_{2n+1}H series have been investigated experimentally in laboratory studies. Their electronic absorption spectra have been recorded in 5 K neon matrices, providing origin band transitions as well as transitions to vibrationally excited states in the $\widetilde{A}^{3}\Sigma_{u}^{-}$ state.^[9] Guided by these data, the gas phase spectra have been recorded for HC_{2n+1}H (n=2-6), using different spectroscopic techniques such as two-color twophoton-ionization^[10] and cavity ring-down spectroscopy.^[11,12] The latter technique has resulted in rovibronically unresolved spectra of $HC_{2n+1}H$ species and their corresponding partially or fully deuterated isotopologues using a pinhole plasma expansion. In more recent work^[13], partially resolved spectra for HC_7H have been reported, using a planar plasma source with a reduced Doppler broadening in combination with an improved laser bandwidth. The present study utilizes the same method and extends results to HC_7D and DC_7D , whose band origin positions have been reported previously,^[12] but a rotational analysis is lacking so far.

The aim of the present study is to enlarge the spectroscopic database for carbon-based radicals to be used as a reference for the identification of species in dilute and optically transparent interstellar clouds.

Experimental

The gas-phase electronic spectra of linear HC₇D and DC₇D are recorded by pulsed cavity ring-down spectroscopy. The experimental setup, shown in Fig. 1, has been documented in Refs.^[13–15], and here some essential details are described. The partially and fully deuterated carbon chain radicals are generated by discharging a pulsed gas mixture $(0.15\% C_2D_2)$ or $\sim 0.35\%$ C₂H₂ + 0.15% C₂D₂ in a 7:3 He:Ar mixture) of 1 ms duration in the throat of a slit discharge nozzle. It is found in the present experiment that, with the same discharge condition, the carbon chain production efficiency can be increased by a factor of 3–4 by adding argon to the expanding gas mixture. This may be due to the fact that argon decreases the overall expanding velocity and consequently results in more collisions in the discharge area. The gas is expanded with a backing pressure of about 10 bar through a long and narrow $(3 \text{ cm} \times 300 \,\mu\text{m})$ slit that is positioned parallel to the laser beam and offset by a few millimeters with respect to

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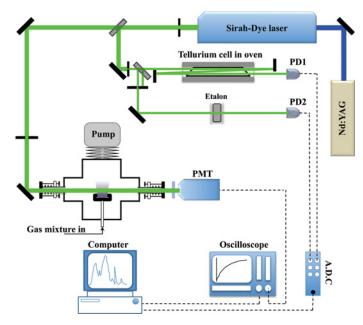


Fig. 1. The experimental setup.

the optical axis of a high finesse cavity of length 58 cm, consisting of two high reflectivity mirrors. Tunable light with a bandwidth of ~ 0.035 cm^{-1} is generated by a tripled Nd:YAG laser (355 nm, 8 Hz) pumped dye laser (Sirah, Cobra-Stretch), operated in a second-order grating configuration, and spatially filtered before it is focused into the optical cavity. Light leaking out of the cavity is detected by a photo-multiplier tube, and the ring-down signals are converted into an optical absorption spectrum.^[16] The laser frequency is accurately calibrated by simultaneously recording a 650°C tellurium vapor absorption spectrum and linearized using an etalon with a free spectral range of 0.7 cm^{-1} . This yields an absolute laser frequency accuracy of 0.02 cm^{-1} .

Results and Discussion

HC_7D

Panel (b) of Fig. 2 shows the experimental spectrum of the $\widetilde{A}^{3}\Sigma_{u}^{-} - \widetilde{X}^{3}\Sigma_{g}^{-}$ electronic origin band transition of HC₇D. The observed band, recorded at a S/N of \sim 25, exhibits a partially resolved rotational structure that is spectrally polluted by overlapping narrow features due to other (smaller) hydrocarbon compounds. This is illustrated in panel (c) for expansion conditions not in favor of carbon chain formation - lower backing pressures and lower discharge current - from which the polluting peaks can be easily discriminated. The spacing between individual rotational transitions in P- and R-branches of the asymmetric HC₇D molecule is not similar to the symmetric HC₇H (or DC₇D) species, as spin-statistics do not apply; i.e., the spacing between subsequent rotational transitions is approximately 2B, rather than 4B. A zoom-in view of the spectrum (panel (a) of Fig. 2) illustrates that many of the individual transitions indeed are resolved.

For a ${}^{3}\Sigma$ electronic state, due to spin-rotational interactions, rotational levels are split into triplet fine structure

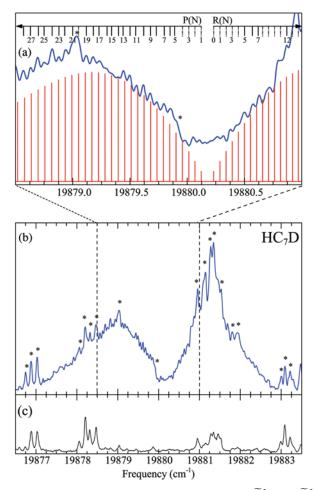


Fig. 2. (a) The expanded spectrum of the $\tilde{A}^3 \Sigma_u^- - \tilde{X}^3 \Sigma_g^-$ electronic transition of HC₇D (upper trace) in the band origin region compared with a simulated stick diagram using an effective ${}^1\Sigma - {}^1\Sigma$ Hamiltonian. The full spectrum is shown in panel (b). Peaks due to overlapping spectra originating from other species are marked by an asterisk and shown in panel (c) for expansion conditions not in favor of carbon chain formation.

levels: F_1 for N = J - 1, F_2 for N = J, and F_3 for N = J + 1. In the general case the energy eigenvalues of triplet fine structure levels of the same N are close to each other and N is a good quantum number for assigning rotational transitions. Under this assumption the rotational analysis of the spectrum can be performed by employing an effective ${}^{1}\Sigma^{-1}\Sigma$ transition, with the assumption that spin-rotation and spin-spin interaction are not affecting the overall pattern too much. Under this assumption of spin-interactions set to zero, simulations are performed with PGopher software.^[17] Since ground-state constants have not been reported for HC7D and DC_7D , the spectrum is analyzed starting from estimated values of B''_0 and B'_0 , assuming that B'_e/B'_0 and B''_e/B''_0 ratios have similar values as for HC₇H in Ref.^[13] In addition, the value of the rotational groundstate constant of HC7D, $B_{e}^{\prime\prime} = 0.02706 \text{ cm}^{-1}$ has been calculated at B3LYP/6-311+ +G^{**} level using GAUSSIAN 03 software.^[18] Guided by the simulated spectrum, in total 24 P-branch and 5 R-branch transitions are identified in the $\widetilde{A}^{3}\Sigma_{u}^{-} - \widetilde{X}^{3}\Sigma_{g}^{-}$ origin band of

Table 1. Derived constants for the $\tilde{A}^3 \Sigma_u^- - \tilde{X}^3 \Sigma_g^-$ electronic origin bands of DC₇H and DC₇D, and a comparison with the isoelectronic species HC₆N, DC₆N and HC₇H

	HC ₇ D	DC ₇ D	HC ₇ H		HC ₆ N	DC ₆ N
$T_{00}{}^{a}$	$19880.177(2)^{b}$	19943.184(5) ^b	19817.895(4) ^c	19817.892(2) ^e	21208.60(5) ^f	21282.10(5) ^f
B_0''	$0.027231(1)^{b}$	$0.026216(2)^{b}$	$0.028354(5)^{c}$	$0.0283263(48)^e$	$0.02806299(2)^d$	0.026982^{f}
B'_0	$0.027133(1)^b$	$0.026112(9)^b$	$0.028221(4)^{c}$	$0.0282298(46)^{e}$	$0.02792(5)^{f}$	$0.02682(5)^{f}$
$D^{''} imes 10^8$	0.0343^{d}	0.0343^{d}	0.0343^{d}	$22.17(39)^{e}$	$0.0343(17)^d$	_
$D' \times 10^8$	0.0343^{d}	0.0343^{d}	0.0343^{d}	$28.12(36)^{e}$		
$B_0^{\prime\prime}/B_0^\prime$	1.0036	1.0039	1.0011	1.0034	1.0051	1.006

All values are in cm^{-1} .

"The uncertainty in the band origin (T_{00}) represents the statistical error as obtained from the least-squares fit. The absolute laser frequency can be determined with a precision of $0.02 \,\mathrm{cm}^{-1}$.

^bThis work.

^cDerived from a remeasured spectrum in this work.

^dRef.^[19].

^eRef.^[19]. Effective constants only.

^fRef.^[20].

HC₇D. These transitions fully reproduce during independent scans. The unambiguous assignment of P(N) lines for $N \le 13$ in the best part of the spectrum, devoid of spectral lines pertaining to polluting species, warrants an accurate determination of the band origin for HC₇D.

The measured frequencies are used in the final leastsquares fit to derive the value of molecular parameters of HC₇D. For this, the band origin (T_{00}) and rotational constants (B'' and B') are fitted simultaneously. Inclusion of distortion constants, D'' and D' (fixed to the values derived for HC₆N in Ref.^[19]) effectively improves the quality of the fit. The resulting parameters are listed in Table 1. Figure 2 shows the simulated stick diagram using these values. The simulation yields a rotational temperature of ~26 K for HC₇D in the slit jet expansion. The resulting ground-state rotational constant, $B''_{e} = 0.02706$ cm⁻¹, is close to the calculated value of $B''_{e} = 0.02706$ cm⁻¹. The origin band of the $\tilde{A}^{3}\Sigma_{u}^{-} - \tilde{X}^{3}\Sigma_{g}^{-}$ electronic transition of HC₇D at 19880.18(2) cm⁻¹ is in agreement with the value reported in Ref.^[12]

DC_7D

The $\widetilde{A}^{3}\Sigma_{u}^{-} - \widetilde{X}^{3}\Sigma_{g}^{-}$ origin band spectrum of DC₇D, recorded at a S/N of ~15, is shown in panel (a) of Fig. 3. As for HC₇D, the spectrum is blended by a number of overlapping narrow peaks that are reproduced for conditions of lower density and discharged current, not in favor of carbon chain formation, as shown in panel (b). The signal-to-noise ratio of the DC_7D spectrum is less than found for HC_7D , but some individual transitions can be resolved and unambiguously resolved, in particular in the P-branch, therewith defining the position of the band origin. As stated before the effective spacing between strong transitions now amounts to about 4B. The rotational analysis of the data is carried out following a similar procedure as described for HC₇D. In total 14 P-branch transitions and 7 R-branch transitions have been assigned. The resulting parameters from a least-squares fit are listed in Table 1. The value for the ground state rotational constant derived for DC₇D, $B''_0 = 0.026216 \text{ cm}^{-1}$, is close to the calculated value of $B''_e = 0.02605 \text{ cm}^{-1}$. Figure 3 shows the simulated stick diagram of individual rotational transitions in P- and R-branch using the values listed in Table 1. Nuclear spin statistics are clearly involved now, making the transitions for even N twice stronger than those for odd N values. The band origin is found at 19,943.18(2) cm⁻¹, very close to the value reported in Ref.^[12] From the observed spectrum we estimate a rotational temperature of about 30 K for DC₇D. The ratios of B''_0/B'_0 for HC₇D and DC₇D amount to 1.0036 and 1.0039, respectively, and indicate that the overall chain length slightly increases upon electronic excitation. These values are consistent with the ratio found for HC₇H (1.0034), and are slightly smaller than that found for DC₆N (1.0060) and HC₆N (1.0051).^[13,20]

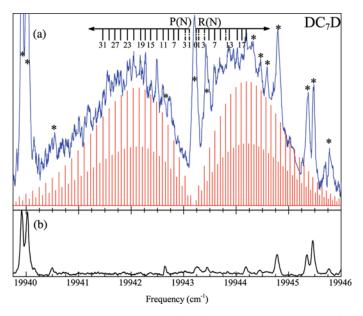


Fig. 3. (a) The observed rotationally resolved spectrum with simulated stick diagram of the $\tilde{A}^{3}\Sigma_{u}^{-} - \tilde{X}^{3}\Sigma_{g}^{-}$ electronic transition of DC₇D. Peaks due to overlapping spectra originating from other species are marked by an asterisk and shown in panel (b) for expansion conditions not in favor of carbon chain formation.

In addition, the spectrum of HC₇H is also remeasured with improved S/N level in the present work; the presently achieved S/N is ~35, while it was ~10 in the previous work.^[13] With the remeasured spectrum, the molecular constants of HC₇H have been improved and the results are summarized in Table 1.

Discussion

Some weak additional features are observed in the experimental spectra of all the three species (HC₇H [13], HC₇D and DC₇D), particularly in the band origin regions where our spectra are not well resolved. This is likely a consequence of fine-structure effects, particularly prominent for transitions involving low rotational levels. For a ${}^{3}\Sigma$ electronic state, the spin-rotation constant, γ is usually very small compared to *B* and even for large *N*-values the resulting splitting of rotational levels will be difficult to resolve. The fine structure for the lowest *J* values is significantly correlated with the value of $\Delta \lambda = \lambda' - \lambda''$. Therefore, the typical observation is an interfering intensity pattern due to overlap of lower *J* rotational (small *N*) levels.

Another possible explanation of the complex spectral structures in the band origin regions is the likely spectral overlapping with other weak vibronic hot bands of the same species. A recent infrared study^[21] on the polyacetylenes $HC_{2n}H$ in a similar plasma expansion has shown that, although a low rotational temperature can be achieved, the mode-dependent vibrational temperatures of HC_4H and HC_6H are found to be up to several hundred kelvin. However, without the knowledge of accurate ground-state spectroscopic parameters, the present experimental spectra do not allow to make a full analysis of these weak features. Future high-resolution infrared measurements on the HC_7H ground state may improve the analysis of the present study.

In previous investigations indication was found for excited state lifetime broadening in HC₇H.^[11,12] We have performed simulations for the presently obtained spectra of the deuterated species by convolving a Lorentzian width to the Gaussian width of 0.04 cm^{-1} , the latter produced by the combined effects of laser line width and Doppler width. An optimum is found for a Lorentzian width of 0.02 cm^{-1} , corresponding to an upper state lifetime of 0.3 ns. This lifetime, resulting from spectra of slightly higher quality, is somewhat larger than the indicative value of 0.1 ns reported by Ball et al.^[12], but in view of the uncertainties associated with the overlapping structures still in reasonable agreement. It proves that nonradiative decay process occur for all three isotopologues.

Conclusion

The present spectroscopic study on HC₇D and DC₇D radicals marks the limit of what can be achieved in terms of resolving rotational manifolds of chains of carbon-based hydrocarbons of increasing length under conditions of slit-jet discharge plasma expansions typically used in a number of laboratories to record optical absorption spectra of such species.^[4,12,14,15] Where the rotational structure of C₆H and

 C_6D radicals could be fully resolved in a setup with a generic pulsed cavity ring-down experiment combined with a slit-jet discharge expansion^[15], for the present chains with seven carbon atoms the rotational sequences become overlapped. In addition indication is found of broadening due to internal conversion of electronic excitation in HC₇H and its deute-rated analogs. The presently observed spectra display an improvement over to those of a previous investigation^[12], just sufficient to extract values for the rotational constants of HC₇D and DC₇D, providing insight in a change of chain length upon electronic excitation.

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